

Figure 5

R = 0,5°C/s
 T_q = 1200°C
 (d₁₂₀₀ = 300 μm)
 Former δ-dendrites (D).
 White interdendritic areas (ID).
 (Most of the δ transformed to γ by the peritectic reaction and transformation.)

× 25 400 μm

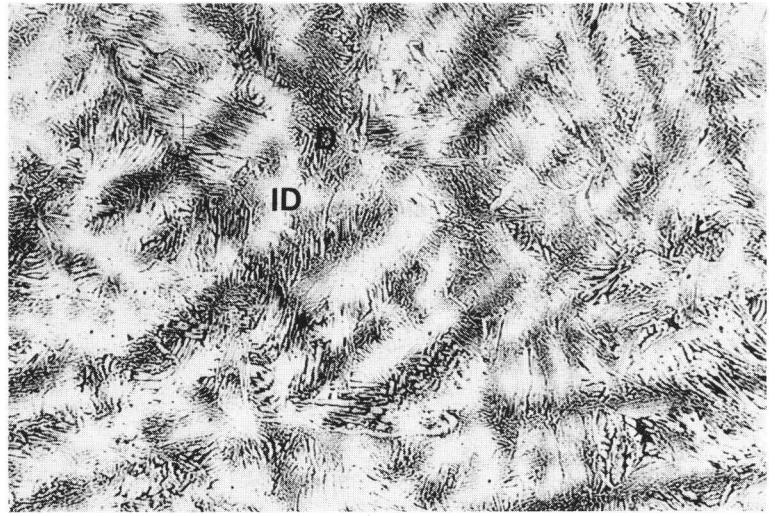


Figure 6

R = 2,0°C/s
 T_q = 1325°C
 13 vol-% dendritic ferrite.
 Figures 6–9: Note that the residual ferrite only appears in the former δ-dendrites.

× 150 100 μm



Figure 7

R = 0,5°C/s
 T_q = 1325°C
 19 vol-% dendritic ferrite (δ).

Figure 8

R = 0,5°C/s
 T_q = 1200°C
 9 vol-% dendritic ferrite (δ).

× 150 100 μm

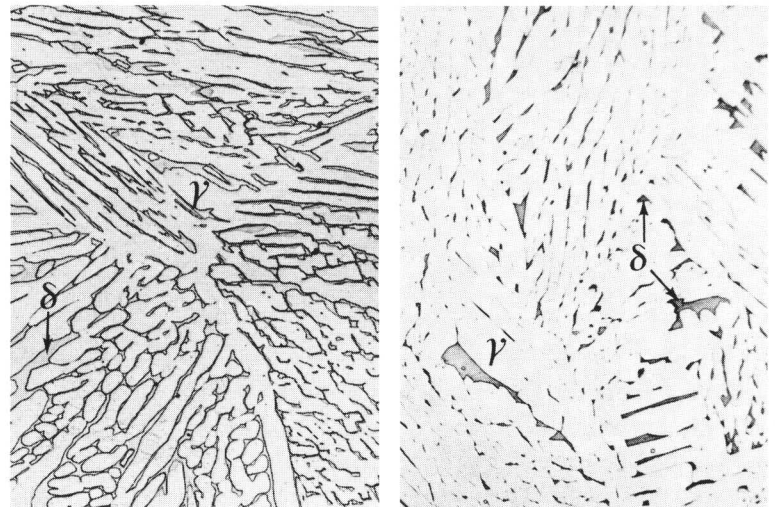
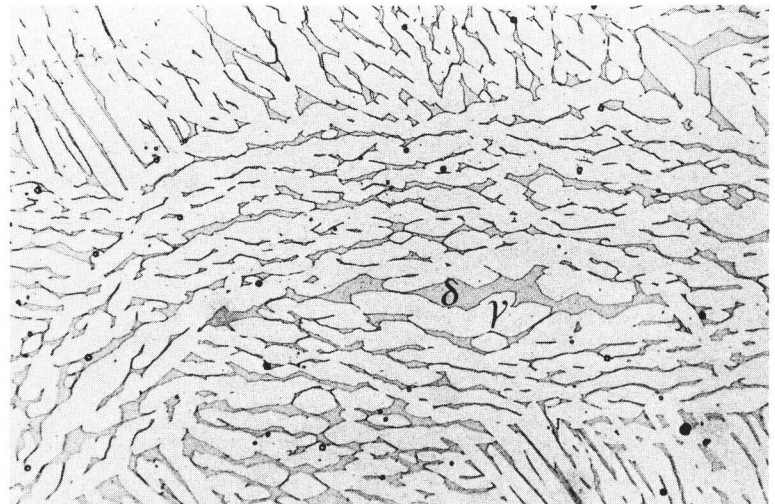


Figure 9

R = 0,1°C/s
 T_q = 1325°C
 20 vol-% dendritic ferrite.

× 150 100 μm



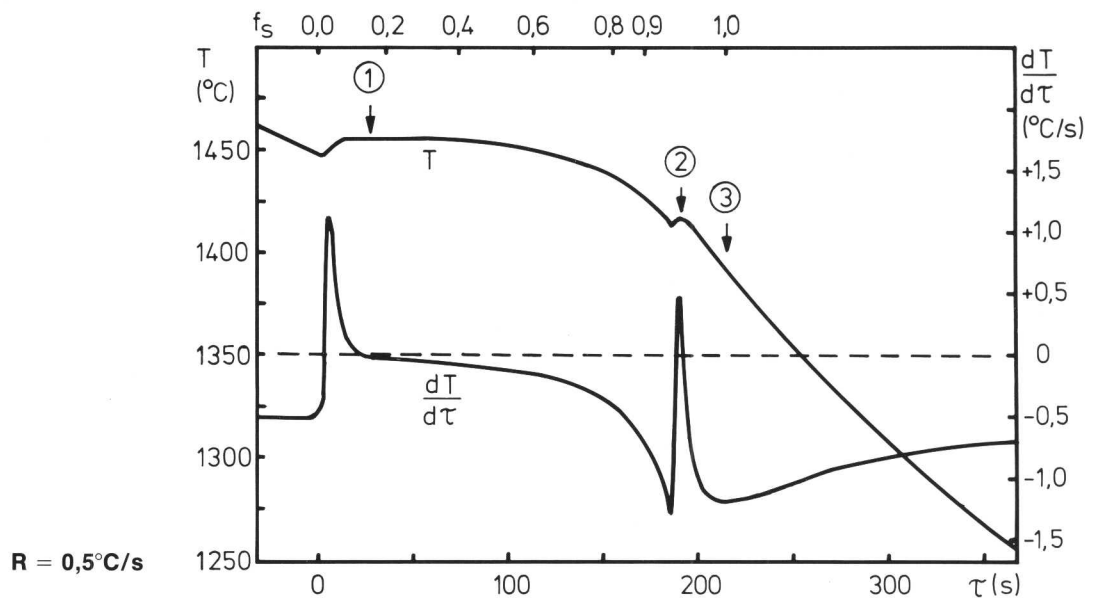
STEEL 403. 0,02 % C 19 % Cr 10 % Ni STAINLESS STEEL**Designations**

SIS	AISI	Werkstoff Nr
(2352)	304L	1.4316

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Al _{tot}	N
0,019	0,31	0,94	0,009	0,010	19,5	10,2	0,11	0,03	0,05	0,002	0,044

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,74$$

Thermal Analysis

	Average Cooling Rate, R, ($^\circ\text{C/s}$)		
	2,0	0,5	0,1
Liquidus temperature, ferritic primary phase, $^\circ\text{C}$ ①	1447	1455	1453
Temperature of austenite formation, $^\circ\text{C}$ ②	1404	1415	1418
Solidus temperature, $^\circ\text{C}$ ③	1365	1390	1405
Solidification range, $^\circ\text{C}$	80	65	50
Solidification time, s	90	220	610
Fraction solidified as ferrite, %	91	92	98

Precipitates**Microsegregation**

Element	Si	Mn	Cr	Ni
I	1,6	1,5	1,1	1,5
P _D			1,2	0,7

$R = 0,5^\circ\text{C/s}$
 $T_q = 1340^\circ\text{C}$

Partly solidified

Figure 1

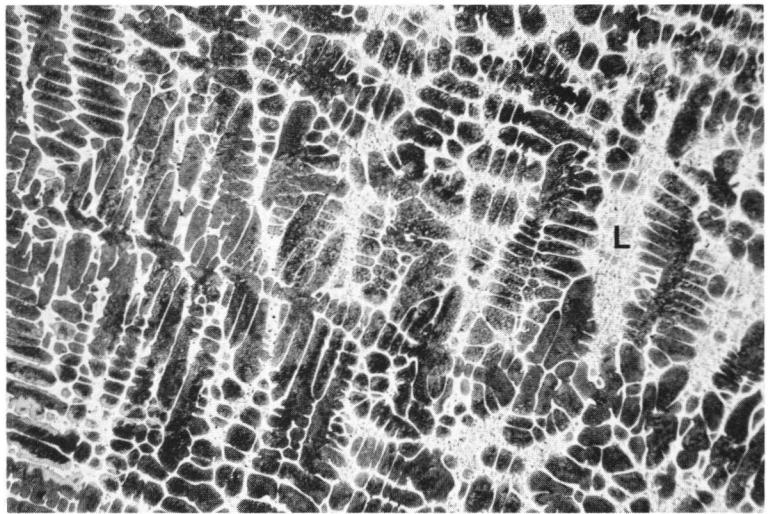
$R = 0,5^{\circ}\text{C/s}$

$T_q = 1450^{\circ}\text{C}$

$d = 65\ \mu\text{m}$

δ -dendrites and quenched liquid (L).

$\times 25$ 



Completely solidified

Figure 2

$R = 2,0^{\circ}\text{C/s}$


$T_q = 1340^{\circ}\text{C}$

$d = 130\ \mu\text{m}$

Figures 2–4: Former δ -dendrites (D).

White interdendritic areas (ID).

(Most of the δ transformed to γ by the peritectic reaction and transformation.)

$\times 25$ 

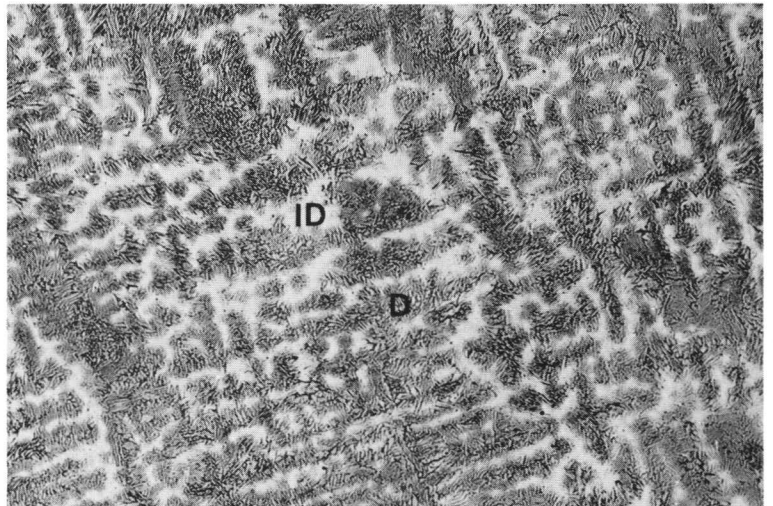


Figure 3

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1340^{\circ}\text{C}$

$d = 160\ \mu\text{m}$


$\times 25$ 




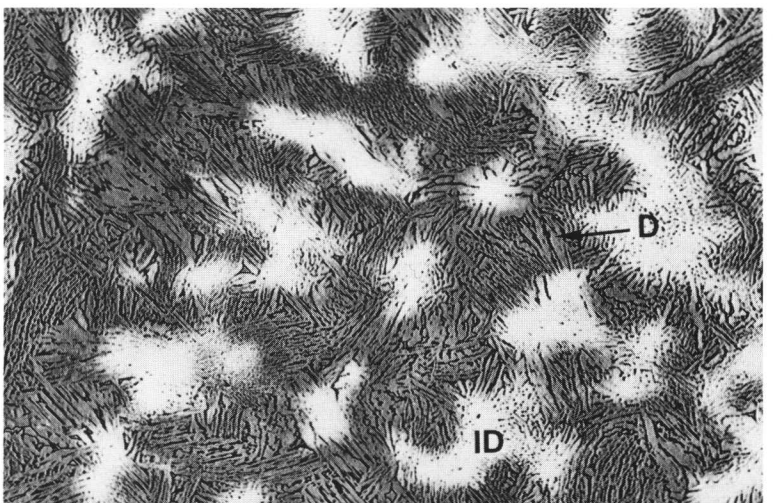
Figure 4

$R = 0,1^{\circ}\text{C/s}$

$T_q = 1340^{\circ}\text{C}$

$d = 500\ \mu\text{m}$

$\times 25$ 



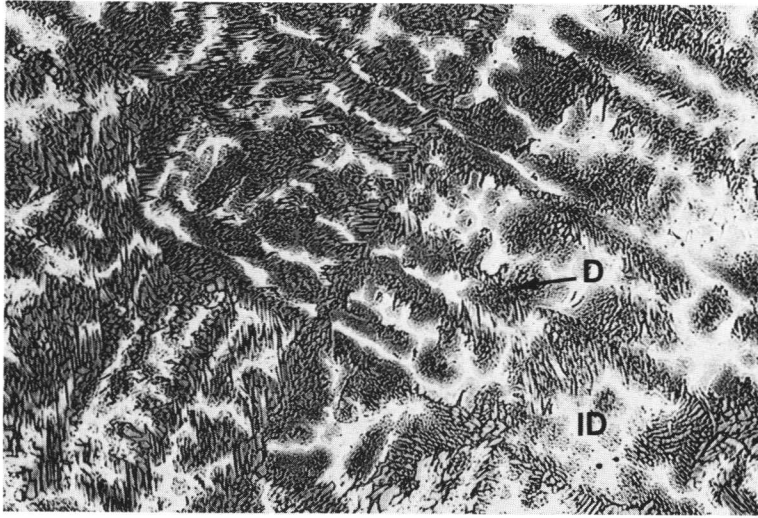


Figure 5

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1200^{\circ}\text{C}$

($d_{1200} = 170 \mu\text{m}$)

Former δ -dendrites (D).

White interdendritic areas (ID).

(Most of the δ transformed to γ by the peritectic reaction and transformation.)

400 $\mu\text{m} \times 25$

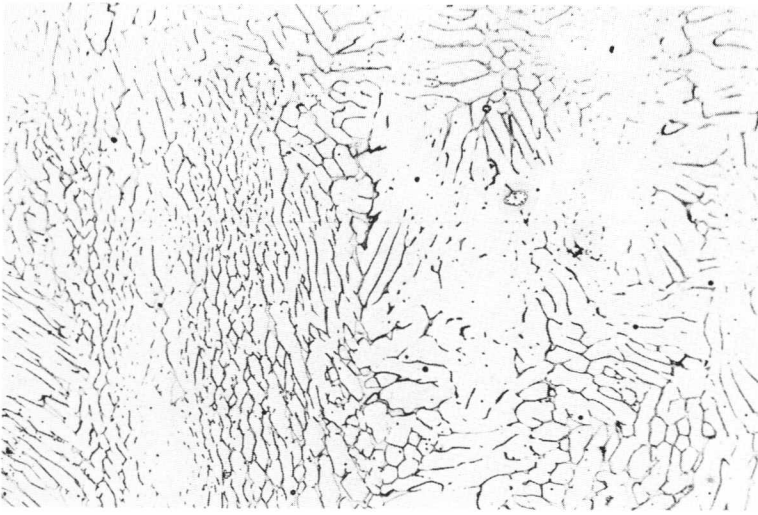


Figure 6

$R = 2,0^{\circ}\text{C/s}$

$T_q = 1340^{\circ}\text{C}$

11 vol-% dendritic ferrite.

Figures 6–9: Note that the residual ferrite only appears in the former δ -dendrites (D).

100 $\mu\text{m} \times 150$

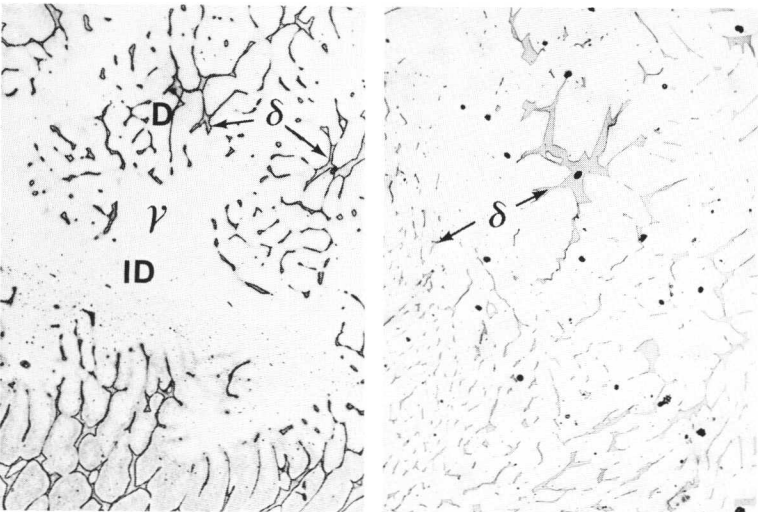


Figure 7

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1340^{\circ}\text{C}$

13 vol-% dendritic

(D) ferrite (δ).

Figure 8

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1200^{\circ}\text{C}$

5,8 vol-% dendritic ferrite.

100 $\mu\text{m} \times 150$

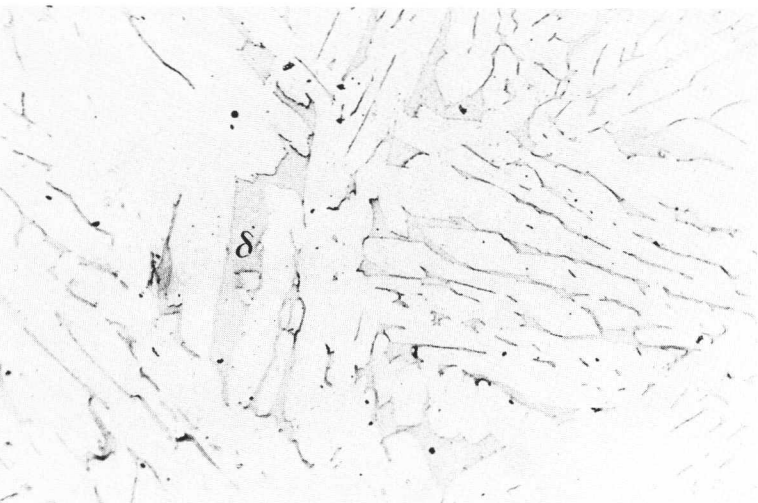


Figure 9

$R = 0,1^{\circ}\text{C/s}$

$T_q = 1340^{\circ}\text{C}$

9 vol-% dendritic ferrite.

100 $\mu\text{m} \times 150$

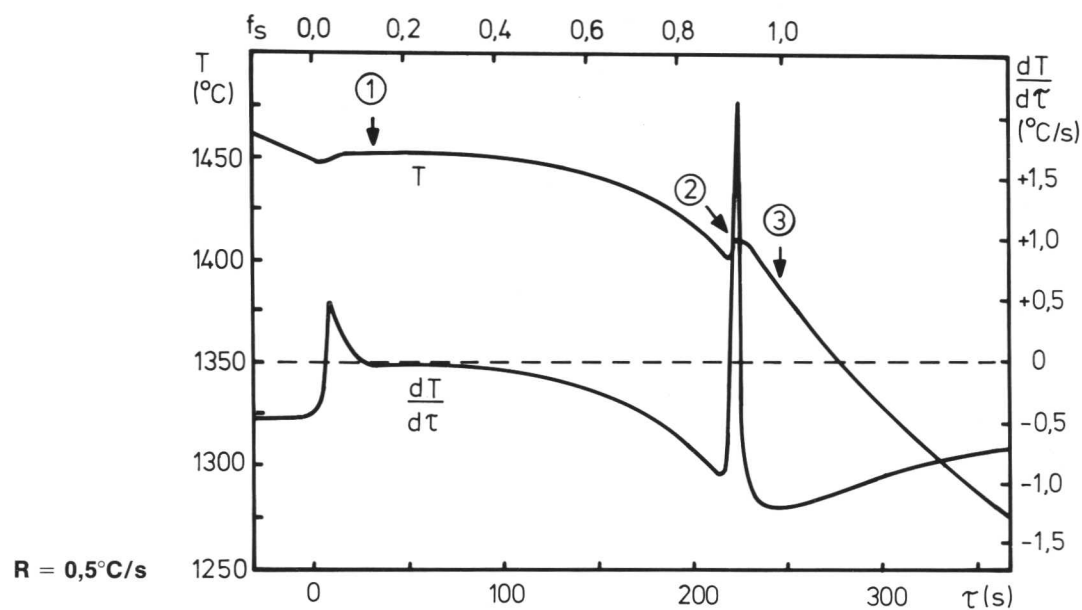
STEEL 404. 0,04 % C 18 % Cr 9 % Ni STAINLESS STEEL**Designations**

SIS	AISI	Werkstoff Nr
2333	304	1.4301

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Al _{tot}	N
0,036	0,44	1,25	0,025	0,010	18,4	9,1	0,38	0,20	0,25	0,002	0,081

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,68$$

Thermal Analysis**Average Cooling Rate, R, ($^{\circ}\text{C/s}$)**

	2,0	0,5	0,1
Liquidus temperature, ferritic primary phase, $^{\circ}\text{C}$ ①	1452	1451	1452
Temperature of austenite formation, $^{\circ}\text{C}$ ②	1423	1409	1424
Solidus temperature, $^{\circ}\text{C}$ ③	1365	1385	1405
Solidification range, $^{\circ}\text{C}$	90	65	50
Solidification time, s	100	250	720
Fraction solidified as ferrite, %	84	82	86

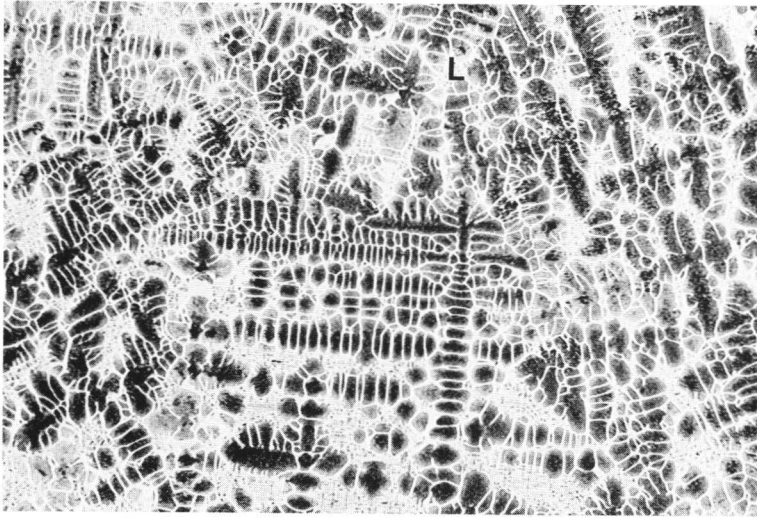
Precipitates

—

Microsegregation

Element	Mn	Cr	Ni
I	1,2	1,1	1,3
P _D		1,2	0,7

$R = 0,5^{\circ}\text{C/s}$
 $T_q = 1340^{\circ}\text{C}$



Partly solidified

Figure 1

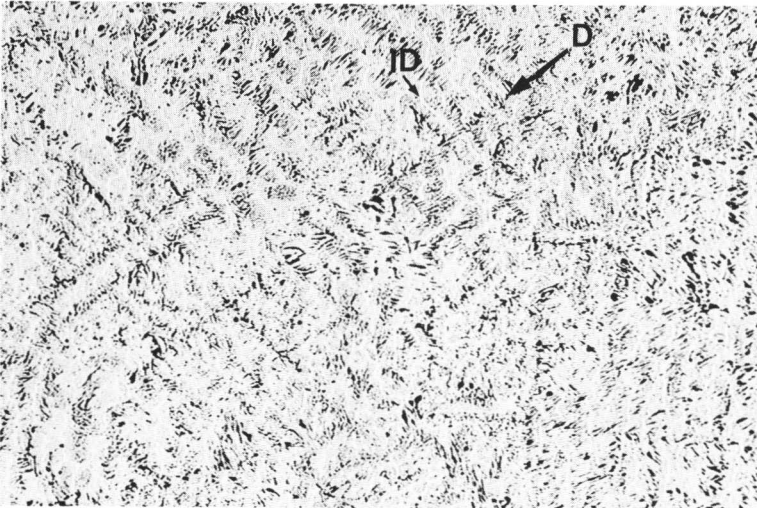
R = 0,5°C/s

T_q = 1447°C

d = 40 μm

δ-dendrites and quenched liquid (L).

400 μm × 25



Completely solidified

Figure 2

R = 2,0°C/s

T_q = 1340°C

d = 125 μm

Figures 2–4: Former δ-dendrites (D).

White interdendritic areas (ID).

(Most of the δ transformed to γ by the peritectic reaction and transformation.)

400 μm × 25

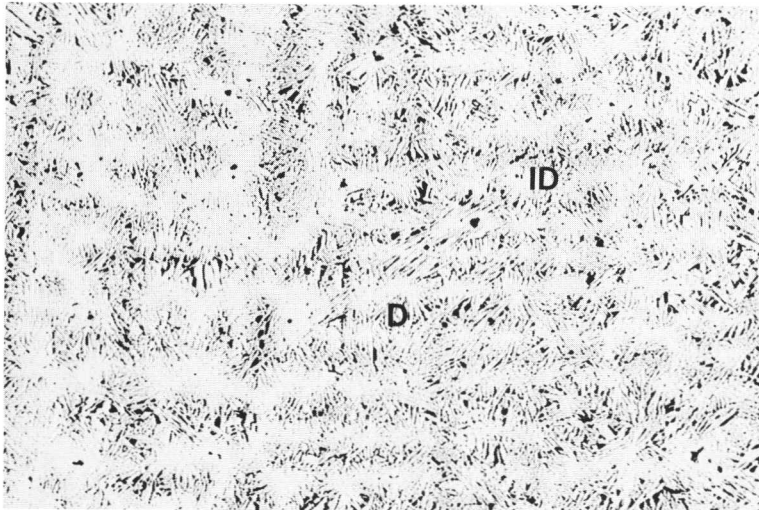


Figure 3

R = 0,5°C/s

T_q = 1340°C

d = 190 μm

400 μm × 25

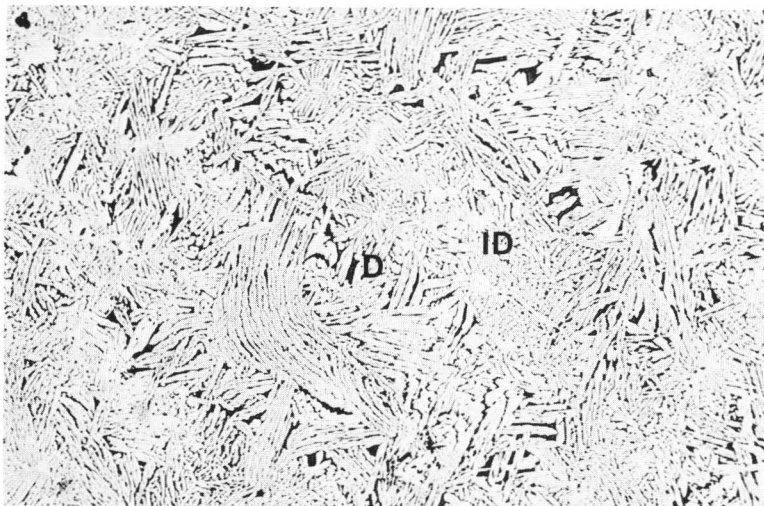


Figure 4

R = 0,1°C/s

T_q = 1340°C

d = 340 μm

400 μm × 25

Figure 5

R = 0,5°C/s
 Tq = 1200°C
 (d₁₂₀₀ = 200μm)
 Former δ-dendrites (D).
 White interdendritic areas (ID).
 (Most of the δ transformed to γ by the peritectic reaction and transformation.)

× 25 400 μm

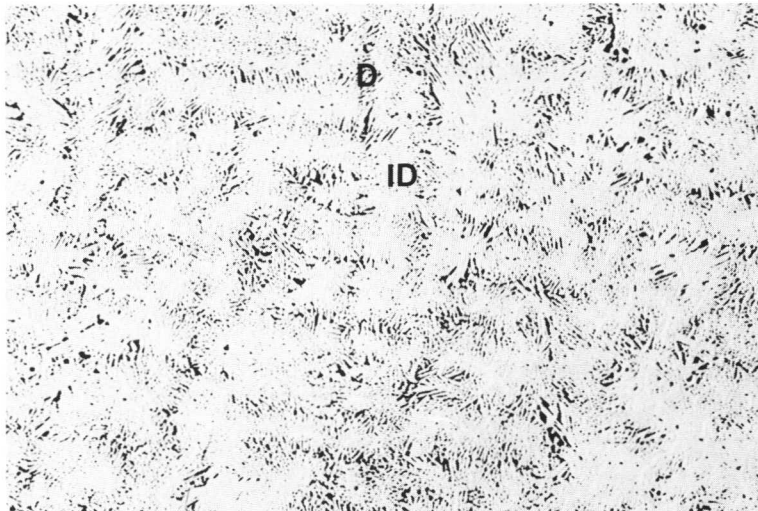


Figure 6

R = 2,0°C/s
 Tq = 1340°C
 2,3 vol-% dendritic ferrite.
 Figures 6–9: Note that the residual ferrite only appears in the former δ-dendrites (D).

× 150 100 μm

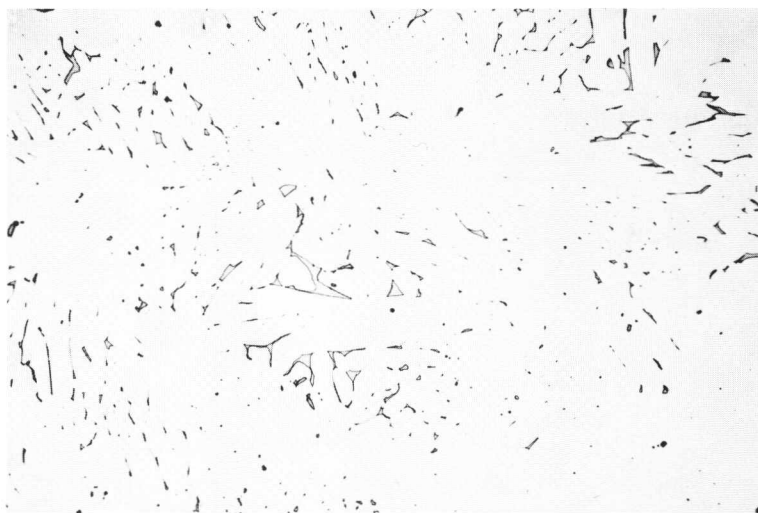


Figure 7

R = 0,5°C/s
 Tq = 1340°C
 4,7 vol-% dendritic ferrite.

Figure 8

R = 0,5°C/s
 Tq = 1200°C
 2,0 vol-% dendritic ferrite.

× 150 100 μm

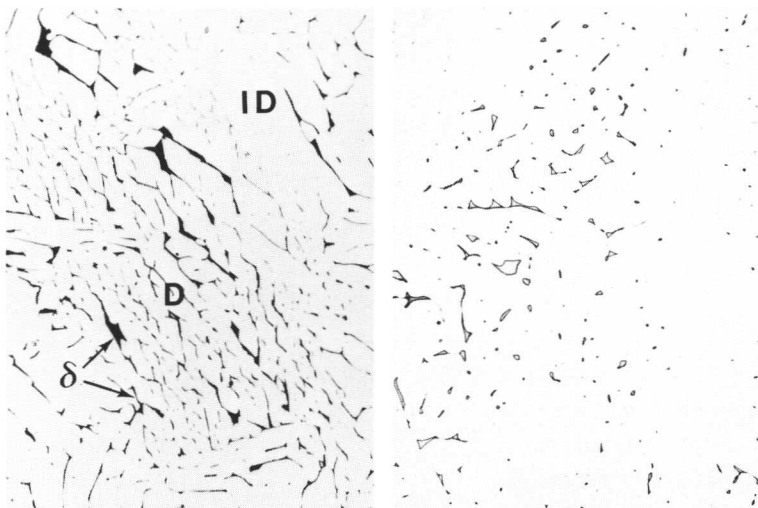
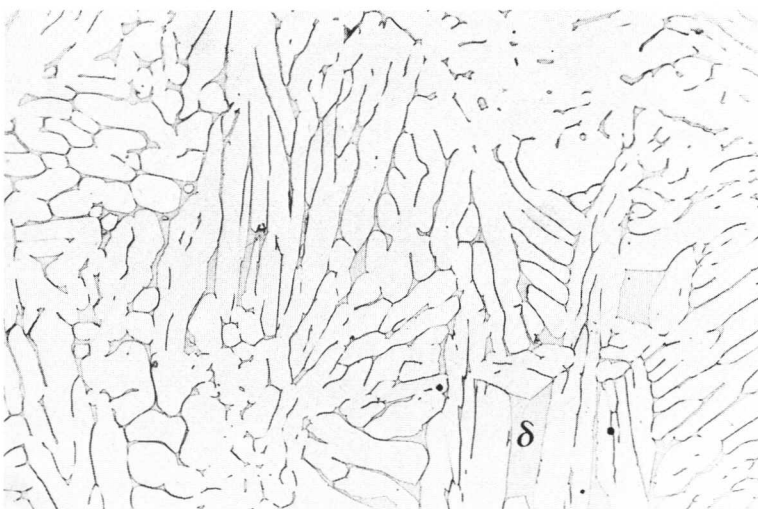


Figure 9

R = 0,1°C/s
 Tq = 1340°C
 10 vol-% dendritic ferrite.

× 150 100 μm



STEEL 405. 0,07 % C 17 % Cr 10 % Ni Ti STAINLESS STEEL

Designations

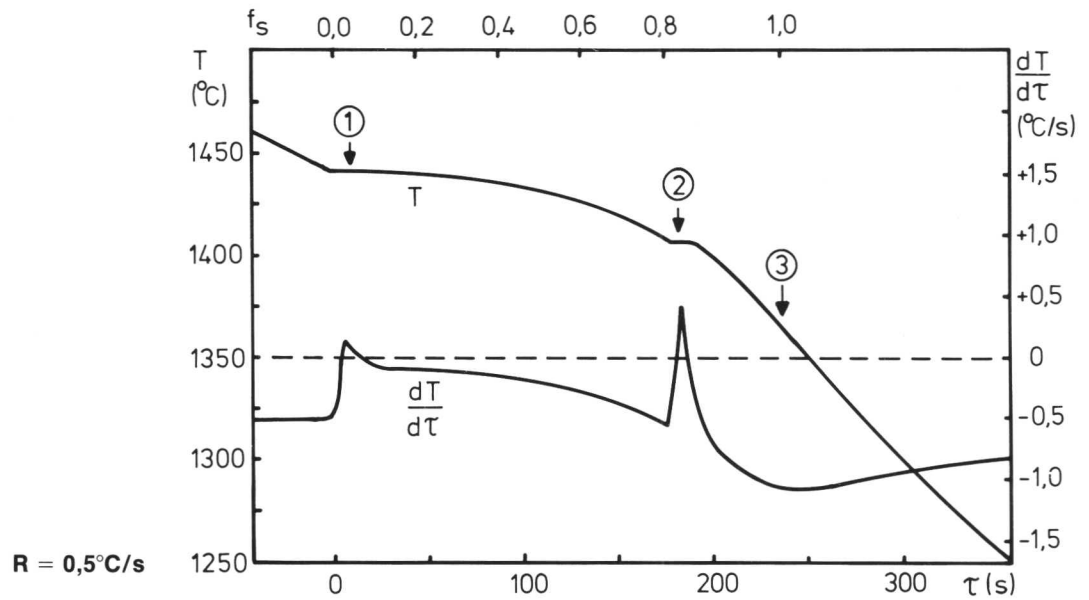
SIS	AISI	Werkstoff Nr
2337	321	1.4541

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Ti	Al _{tot}	N
0,068	0,59	1,44	0,028	0,001	17,2	10,3	0,47	0,24	0,27	0,51	0,048	0,005

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,61$$

Thermal Analysis



	Average Cooling Rate, R , (°C/s)		
	2,0	0,5	0,1
Liquidus temperature, ferritic primary phase, °C (1)	1436	1440	1440
Temperature of austenite formation, °C (2)	1397	1406	1412
Solidus temperature, °C (3)	1335	1370	1390
Solidification range, °C	100	70	50
Solidification time, s	105	235	680
Fraction solidified as ferrite, %	82	82	82

Precipitates

Ti(CN), (see figures 6–8).

Microsegregation

Element	Si	Mn	Cr	Ni
I	1,6	1,6	1,1	1,5
P _D			1,2	0,7

$R = 0,5 \text{ } ^\circ\text{C/s}$
 $T_q = 1320 \text{ } ^\circ\text{C}$

Partly solidified

Figure 1

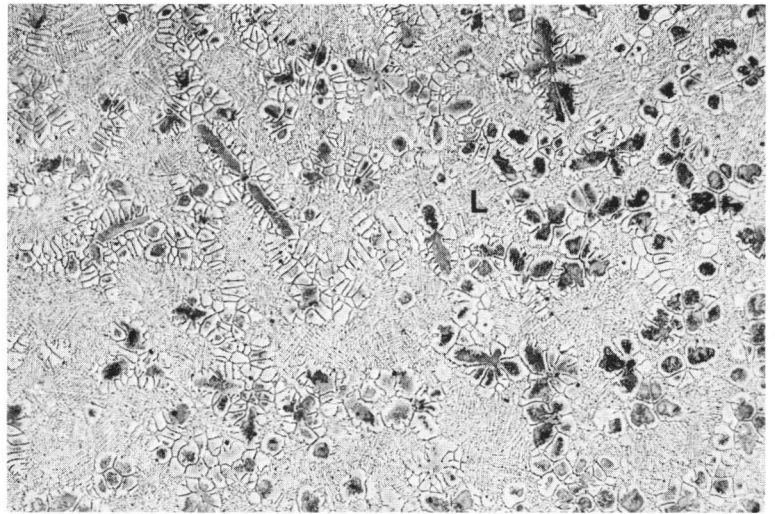
$R = 0,5^{\circ}\text{C}/\text{s}$

$T_q = 1430^{\circ}\text{C}$

$d = 50\ \mu\text{m}$

δ -dendrites and quenched liquid (L).

$\times 25$ 



Completely solidified

Figure 2

$R = 2,0^{\circ}\text{C}/\text{s}$

$T_q = 1320^{\circ}\text{C}$

$d = 85\ \mu\text{m}$

Figures 2–4: Former δ -dendrites (D).

White interdendritic areas (ID).

(Most of the δ transformed to γ by the peritectic reaction and transformation.)

$\times 25$ 

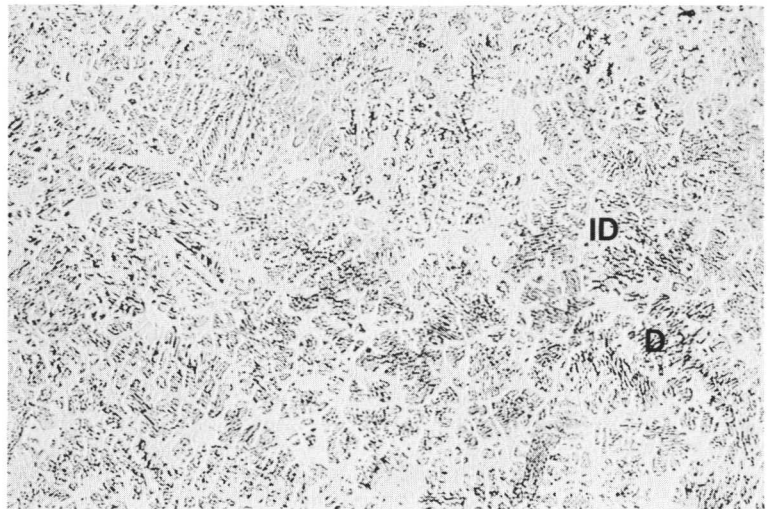


Figure 3

$R = 0,5^{\circ}\text{C}/\text{s}$

$T_q = 1320^{\circ}\text{C}$

$d = 110\ \mu\text{m}$

$\times 25$ 

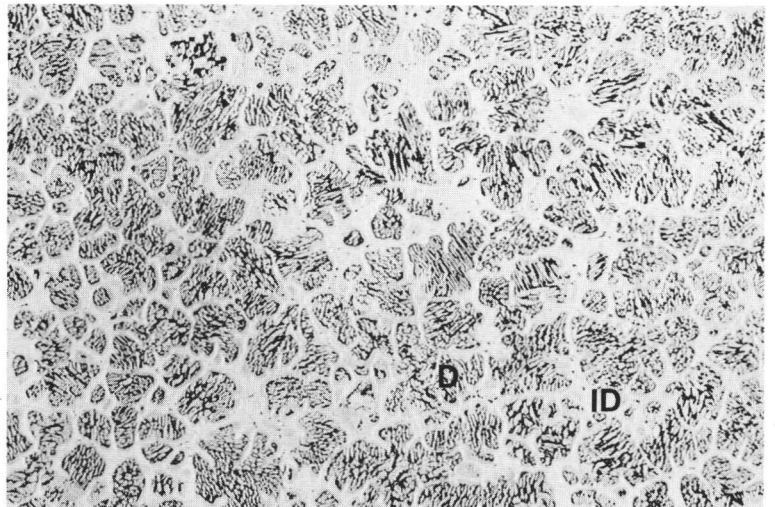


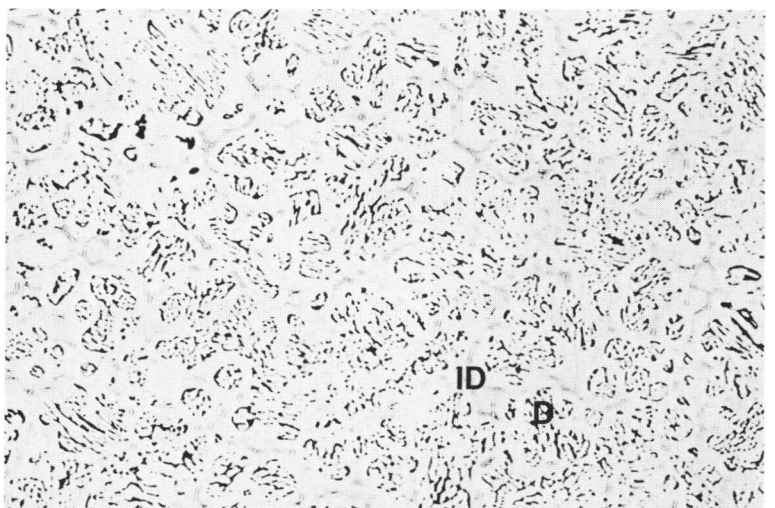
Figure 4

$R = 0,1^{\circ}\text{C}/\text{s}$

$T_q = 1320^{\circ}\text{C}$

$d = 200\ \mu\text{m}$

$\times 25$ 



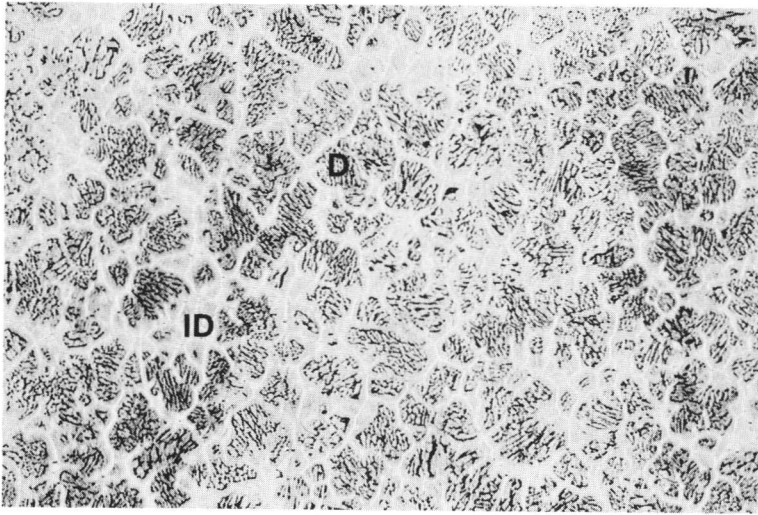


Figure 5

$R = 0,5^\circ\text{C/s}$

$T_q = 1200^\circ\text{C}$

($d_{1200} = 150 \mu\text{m}$)

Former δ -dendrites (D).

White interdendritic areas (ID).

(Most of the δ transformed to γ by the peritectic reaction and transformation.)

400 μm $\times 25$

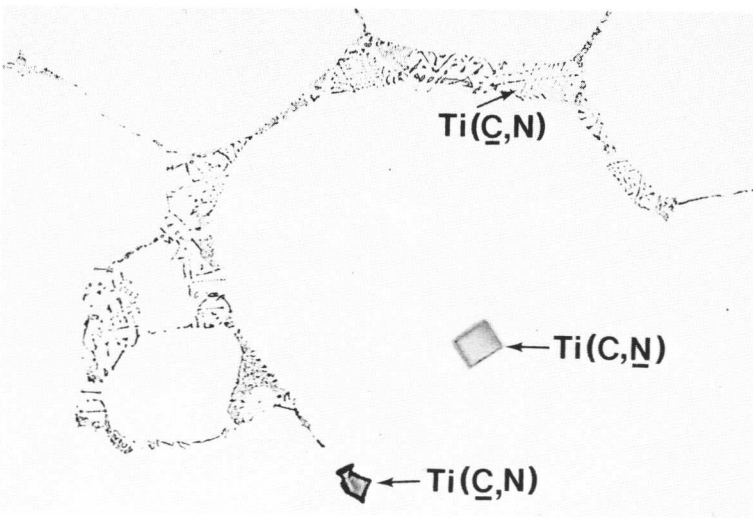


Figure 6

$R = 0,5^\circ\text{C/s}$

$T_q = 1380^\circ\text{C}$

Ti (C,N)

10 μm $\times 1000$

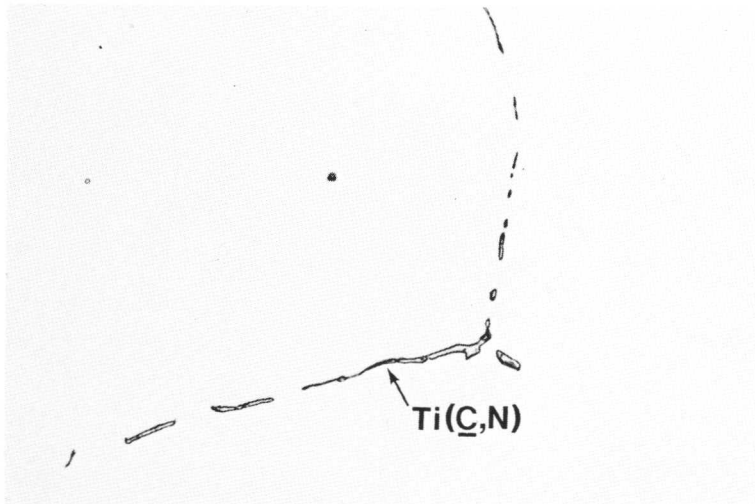


Figure 7

$R = 0,5^\circ\text{C/s}$

$T_q = 1200^\circ\text{C}$

Coalesced eutectic Ti(C,N).

10 μm $\times 1000$

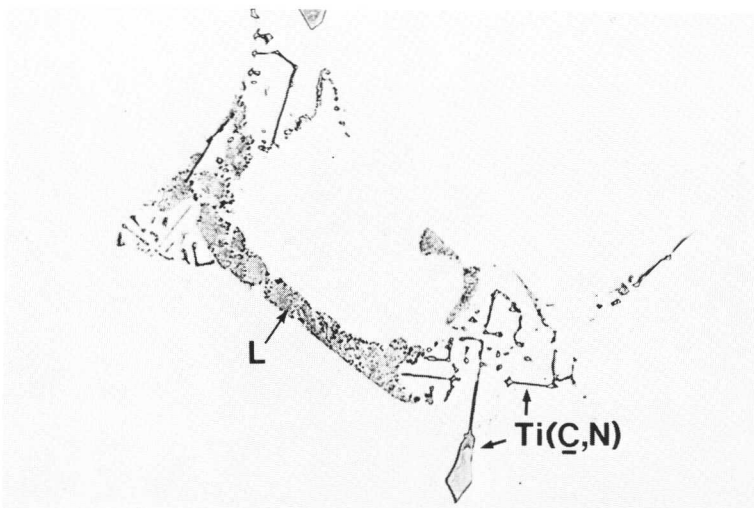


Figure 8

$R = 0,1^\circ\text{C/s}$

$T_q = 1380^\circ\text{C}$

Formation of eutectic Ti(C,N).

($L \rightarrow \text{Ti(C,N)} + \gamma$)

10 μm $\times 1000$

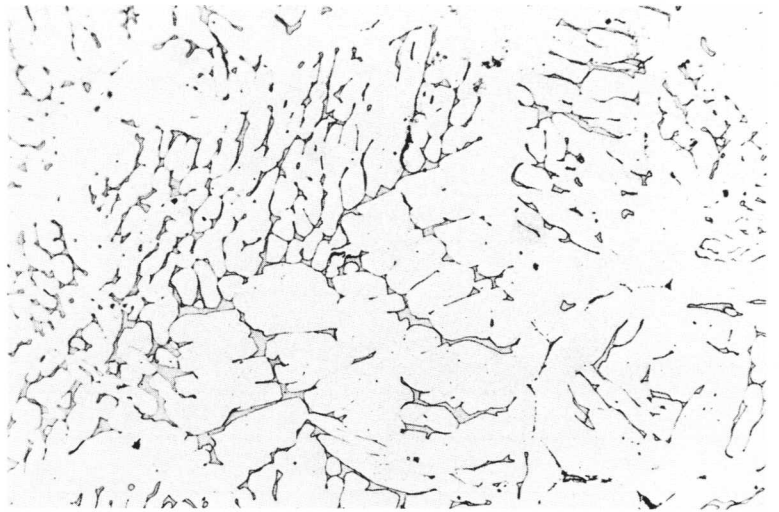
Figure 9

R = 2,0°C/s

T_q = 1320°C

6 vol-% dendritic ferrite.

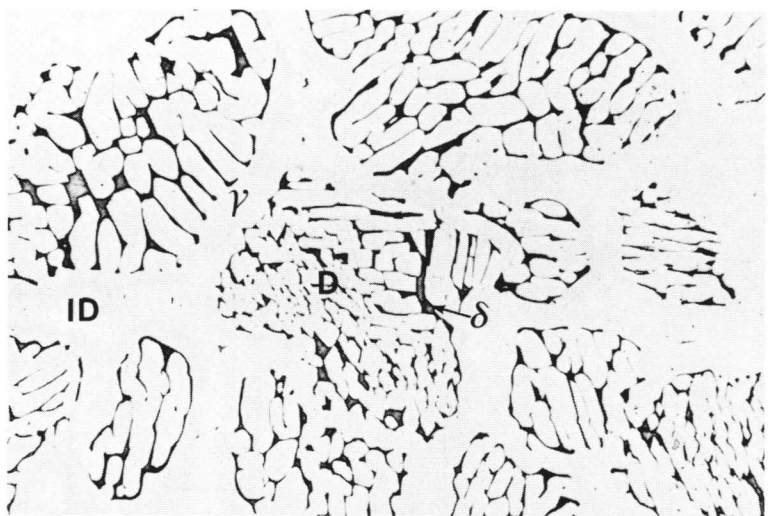
Figures 9–12: Note that the residual ferrite only appears in the former δ-dendrites (D).

× 150 **Figure 10**

R = 0,5°C/s

T_q = 1320°C

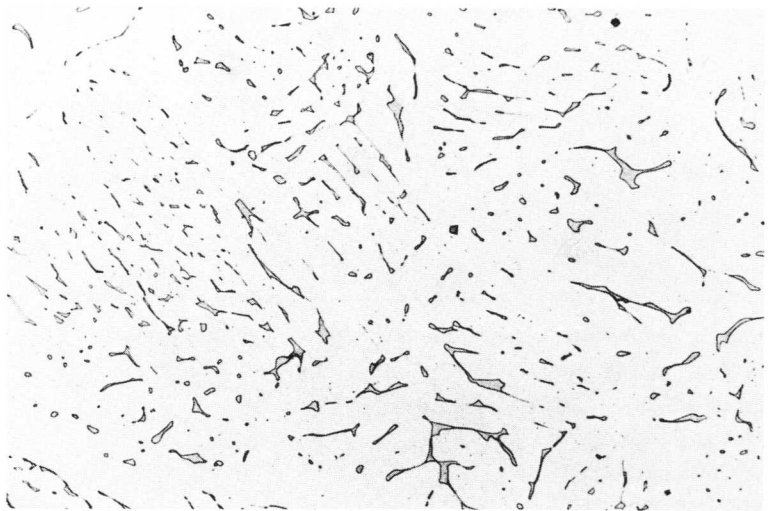
8 vol-% dendritic ferrite.

× 150 **Figure 11**

R = 0,5°C/s

T_q = 1200°C

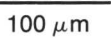
4,1 vol-% dendritic ferrite

× 150 **Figure 12**

R = 0,1°C/s

T_q = 1320°C

4,8 vol-% dendritic ferrite

× 150 

STEEL 406. 0,05 % C 17 % Cr 12 % Ni 2,8 % Mo Nb STAINLESS STEEL

Designations

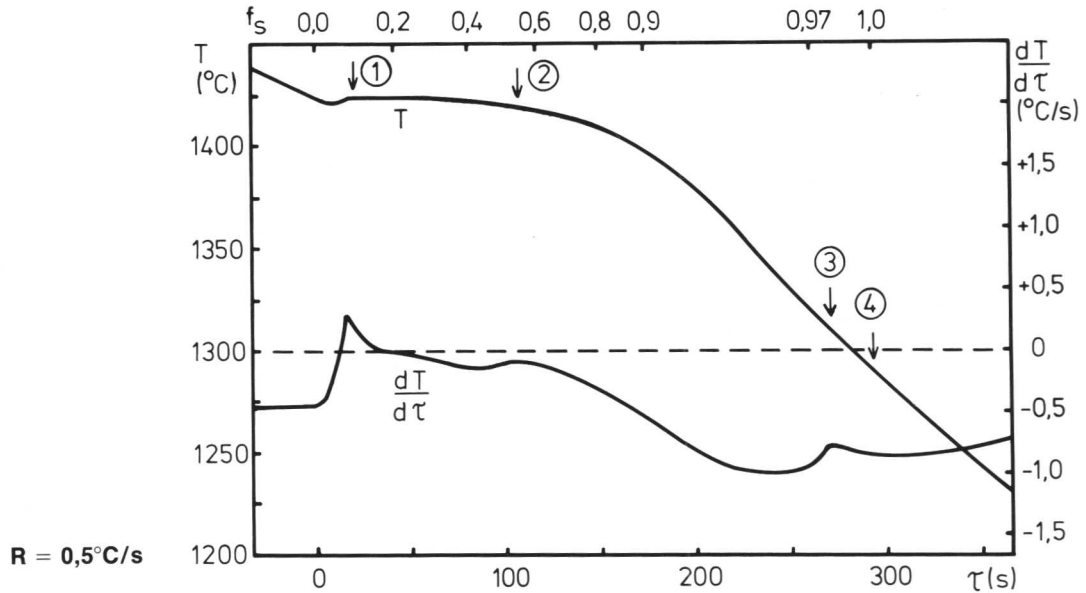
SIS	AISI	Werkstoff Nr
—	316 Cb	1.4583

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Nb	Al _{tot}	N
0,052	0,44	1,71	0,013	0,007	17,2	12,6	2,80	0,03	0,03	0,54	0,004	0,010

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,58$$

Thermal Analysis



	Average Cooling Rate, R, (°C/s)		
	2,0	0,5	0,1
Liquidus temperature, ferritic and austenitic primary phases, °C ①	1420	1423	1424
Temperature of maximum rate of formation of austenite, °C ②	1410	1418	1417
Temperature of carbide formation, °C ③	1330 – 1275	1330 – 1290	1330 – 1305
Solidus temperature, °C ④	1275	1290	1305
Solidification range, °C	145	130	120
Solidification time, s	130	300	1240
Fraction solidified as ferrite, %	<60	<45	<42

Precipitates

1. Interdendritic ferrite, (see figures 4, 6 – 11).
2. Eutectic NbC, (see figures 6, 7, 9 – 11).

Microsegregation

Element	Si	Mn	Cr	Ni
I	1,7	1,5	1,1	1,4
P _D			1,3	0,6
P _{ID}			1,1	0,8

R = 0,5 °C/s
T_q = 1285 °C

Partly solidified

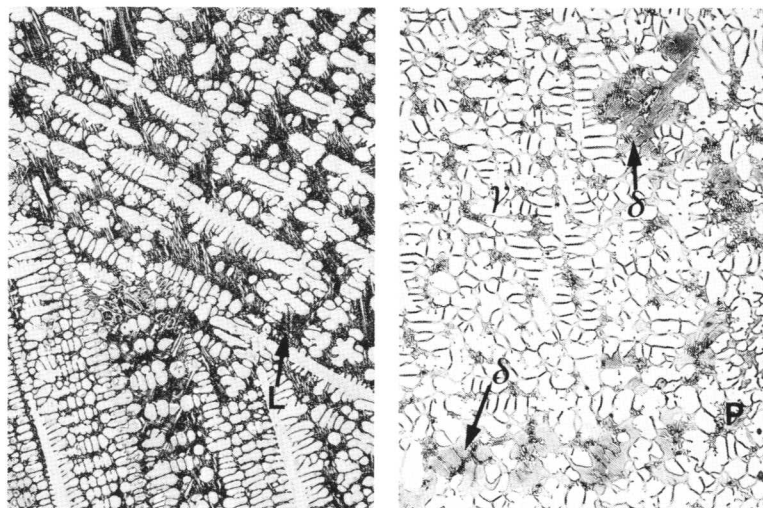
Figure 1a

R = 0,5°C/s
 Tq = 1423°C
 d = 45 μm
 δ-dendrites and quenched liquid (L).

Figure 1b

R = 0,5°C/s
 Tq = 1415°C
 d = 45 μm
 δ-dendrites, γ-dendrites and quenched liquid (L). Peritectic reaction (P).

× 25 400 μm



Completely solidified

Figure 2

R = 2,0°C/s
 Tq = 1270°C
 d = 65 μm
 Figures 2–3: Former δ-dendrites, γ-dendrites, dendritic and interdendritic ferrite, (compare figures 8–9).

× 25 400 μm



Figure 3

R = 0,5°C/s
 Tq = 1285°C
 d = 80 μm

× 25 400 μm

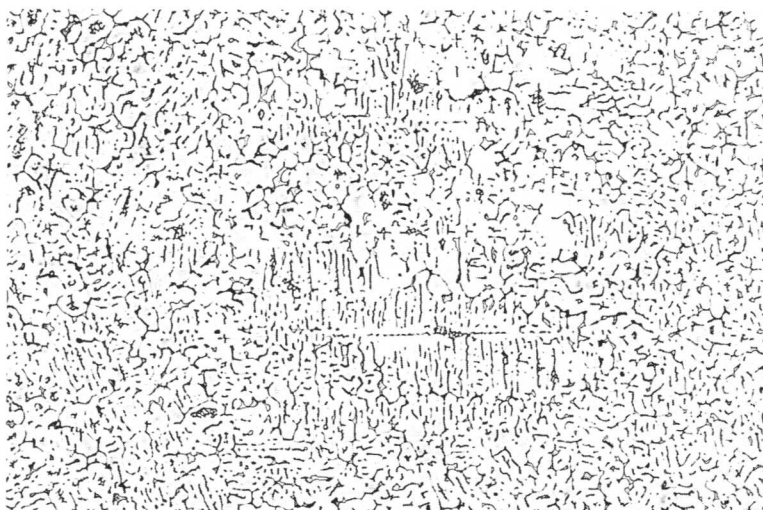
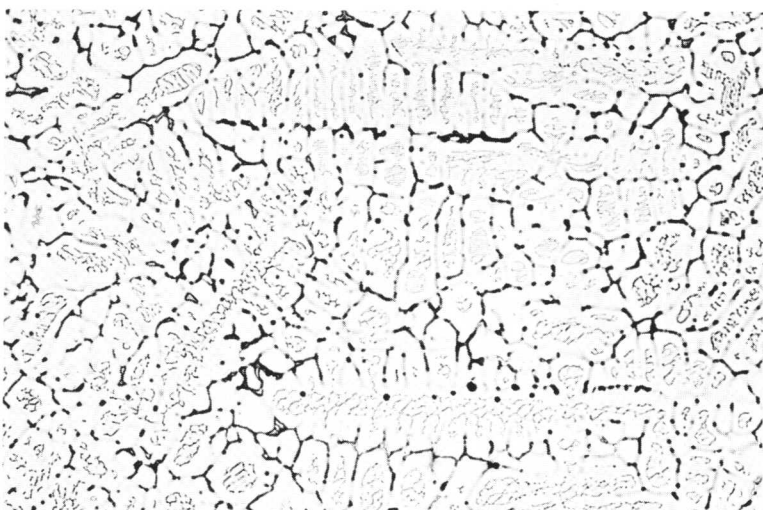


Figure 4

R = 0,1°C/s
 Tq = 1300°C
 d = 135 μm
 Former δ-dendrites, dendritic and interdendritic ferrite, (compare figure 11).

× 25 400 μm



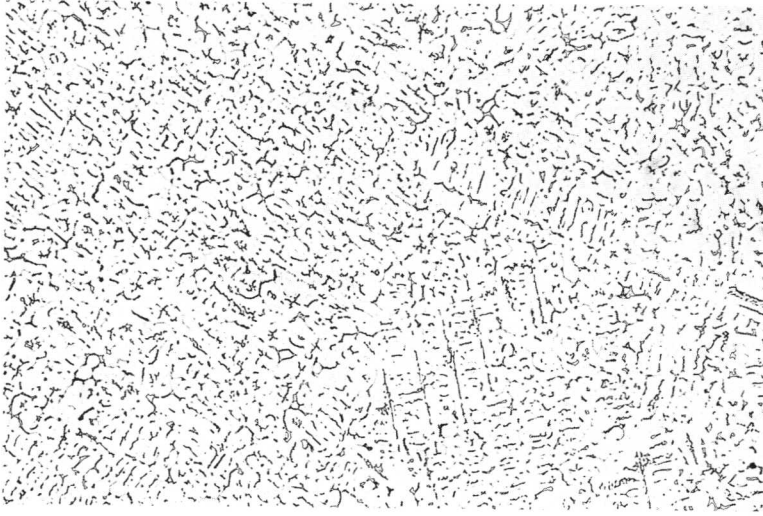


Figure 5

R = 0,5°C/s

T_q = 1200°C

(d₁₂₀₀ = 85μm)

Former δ-dendrites, γ-dendrites, dendritic and interdendritic ferrite, (compare figure 10).

400 μm × 25

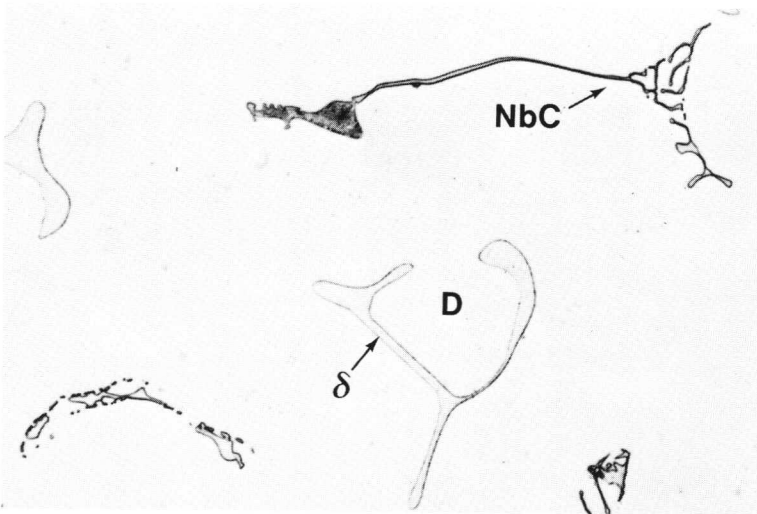


Figure 6

R = 0,5°C/s

T_q = 1200°C

Eutectic NbC and dendritic (D) ferrite (δ).

25 μm × 600

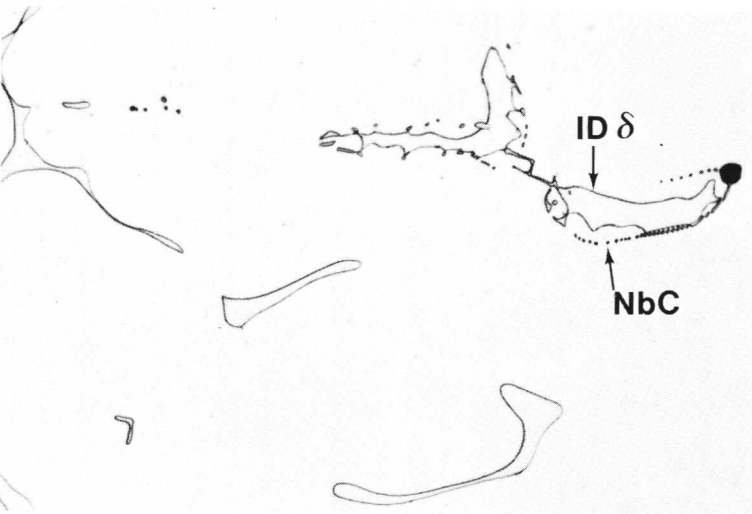


Figure 7

R = 0,5°C/s

T_q = 1200°C

Solid state precipitation of NbC around interdendritic ferrite (IDδ).

25 μm × 600

Figure 8

R = 2,0°C/s
 T_q = 1270°C
 4,0 vol-% ferrite, dendritic (D δ) and interdendritic (ID δ).

× 150 100 μ m

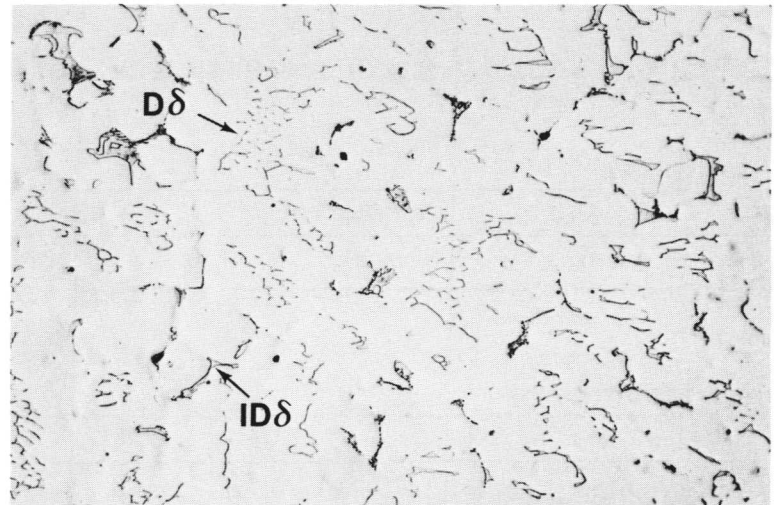


Figure 9

R = 0,5°C/s
 T_q = 1285°C
 4,0 vol-% ferrite, dendritic (D δ) and interdendritic (ID δ).

× 150 100 μ m

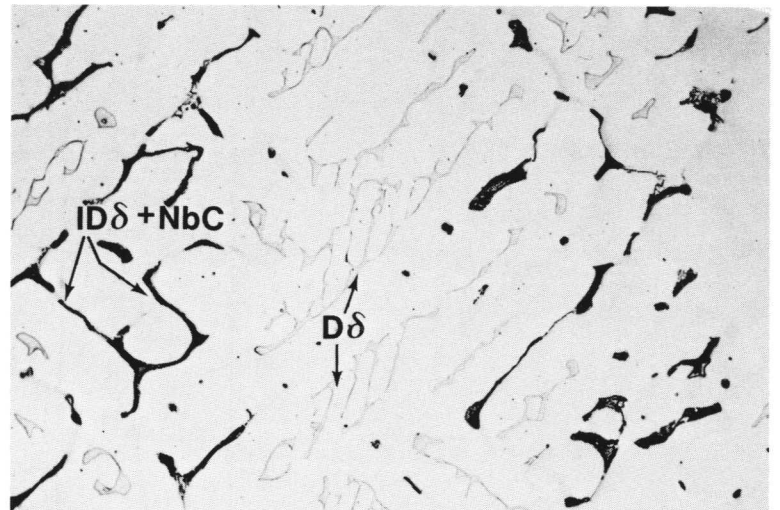


Figure 10

R = 0,5°C/s
 T_q = 1200°C
 2,9 vol-% ferrite, dendritic (D δ) and interdendritic (ID δ).

× 150 100 μ m

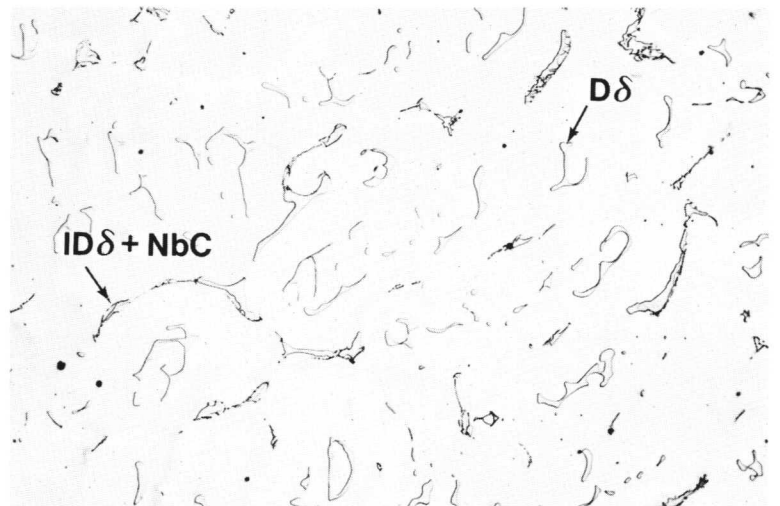
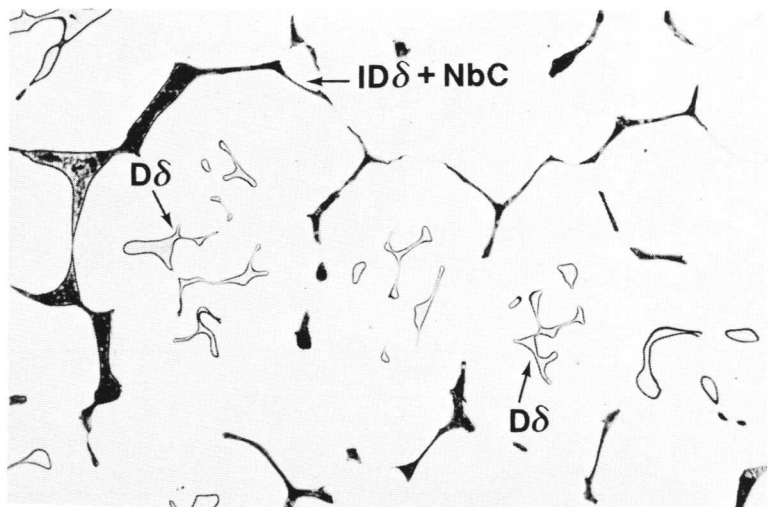


Figure 11

R = 0,1°C/s
 T_q = 1300°C
 3,9 vol-% ferrite, dendritic (D δ) and interdendritic (ID δ).

× 150 100 μ m



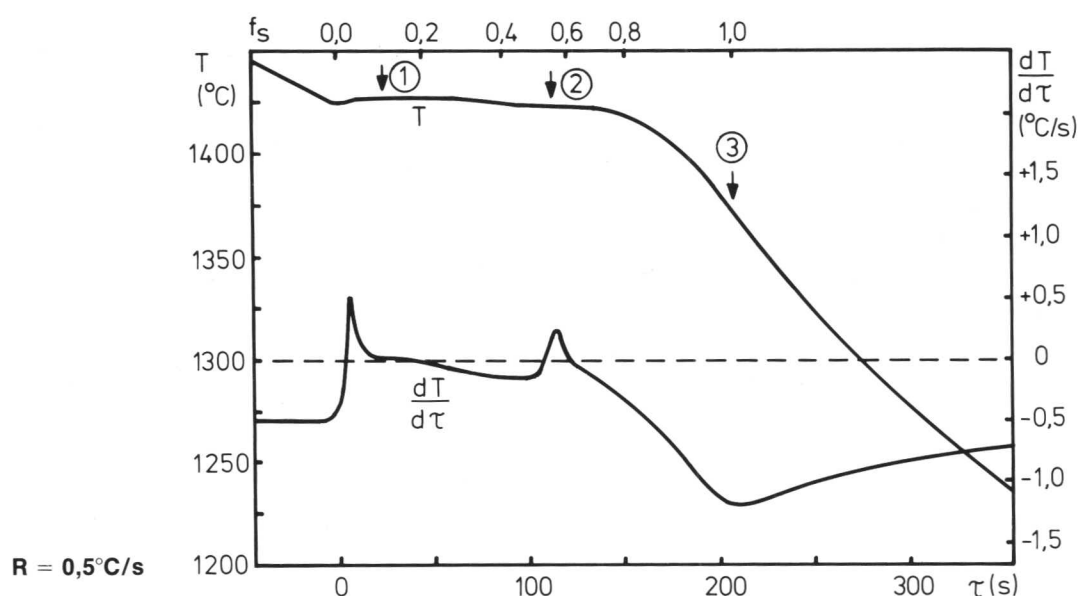
STEEL 407. 0,02% C 17% Cr 13% Ni 2,5% Mo STAINLESS STEEL**Designations**

SIS	AISI	Werkstoff Nr
2353	316 L	1.4435

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Al _{tot}	N
0,023	0,53	1,58	0,020	0,006	17,2	13,5	2,63	0,19	0,07	0,004	0,031

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,43$$

Thermal Analysis

	Average Cooling Rate, R, ($^\circ\text{C/s}$)		
	2,0	0,5	0,1
Liquidus temperature, ferritic and austenitic primary phases, $^\circ\text{C}$ ①	1423	1427	1428
Temperature of maximum rate of formation of austenite, $^\circ\text{C}$ ②	1418	1421	1425
Solidus temperature, $^\circ\text{C}$ ③	1345	1375	1380
Solidification range, $^\circ\text{C}$	80	50	45
Solidification time, s	100	210	660
Fraction solidified as ferrite, %	<46	<50	<34

Precipitates

Interdendritic ferrite, (see figures, 5, 8 – 12).

Microsegregation

Element	Mn	Cr	Ni	Mo
I	1,5	1,2	1,2	2,2
P _{ID}		1,2	0,8	

$R = 0,5^\circ\text{C/s}$
 $T_q = 1320^\circ\text{C}$

Partly solidified

Figure 1

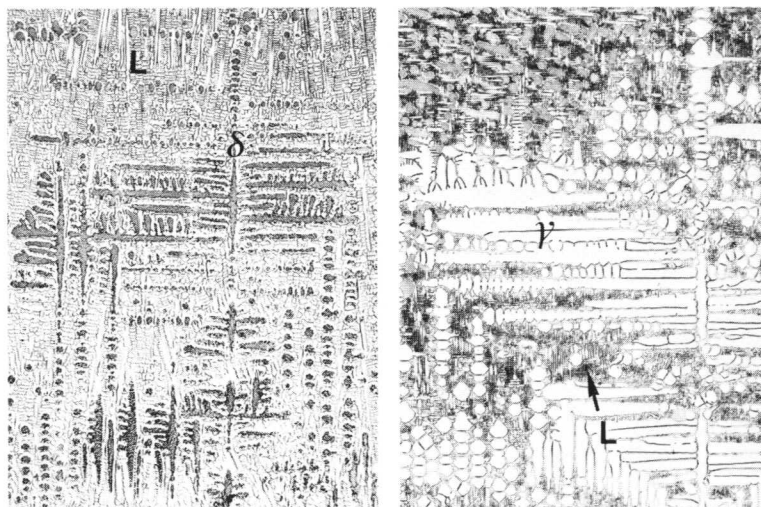
R = 0,5°C/s

T_q = 1427°C

d = 55 μm

δ- and γ- dendrites, growing simultaneously and quenched liquid (L).

× 25 400 μm



Completely solidified

Figure 2

R = 2,0°C/s

T_q = 1320°C

d = 40 μm

Former δ-dendrites, γ-dendrites, dendritic and interdendritic ferrite, (compare figure 9).

× 25 400 μm



Figure 3

R = 0,5°C/s

T_q = 1320°C

d = 90 μm

Figures 3–4: Former δ-dendrites, γ-dendrites and interdendritic ferrite, (compare figures 10 and 12).

× 25 400 μm

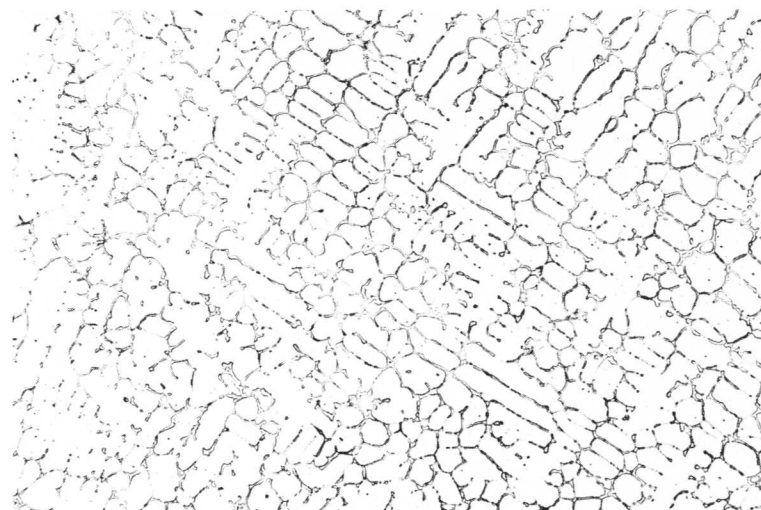


Figure 4

R = 0,1°C/s

T_q = 1320°C

d = 100 μm



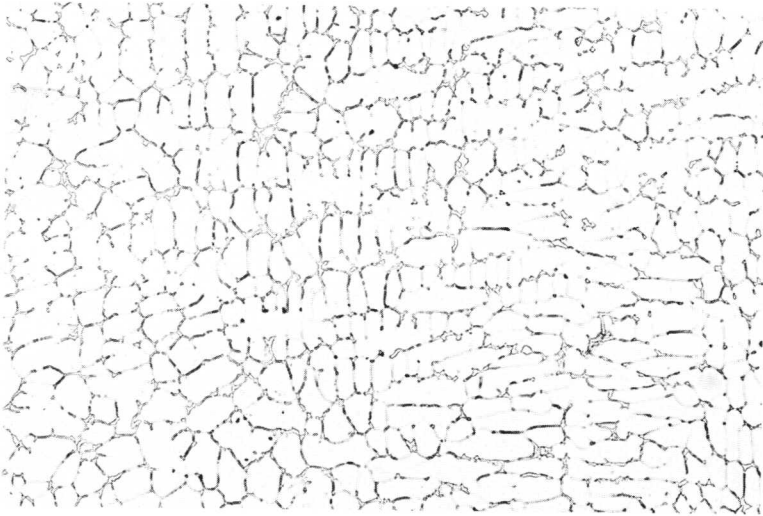


Figure 5

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1200^{\circ}\text{C}$

($d_{1200} = 100\ \mu\text{m}$)

Former δ -dendrites, γ -dendrites and interdendritic ferrite, (compare figure 11).

400 μm $\times 25$

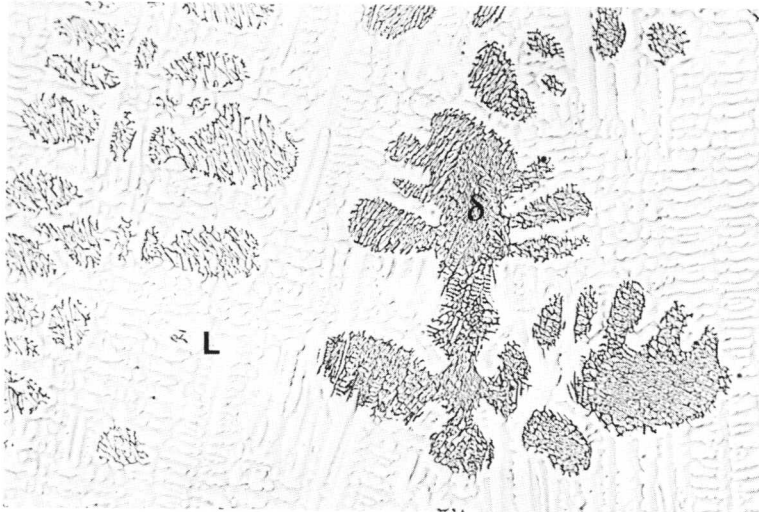


Figure 6

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1427^{\circ}\text{C}$

δ -dendrites and quenched liquid (L), (compare figure 1).

100 μm $\times 150$

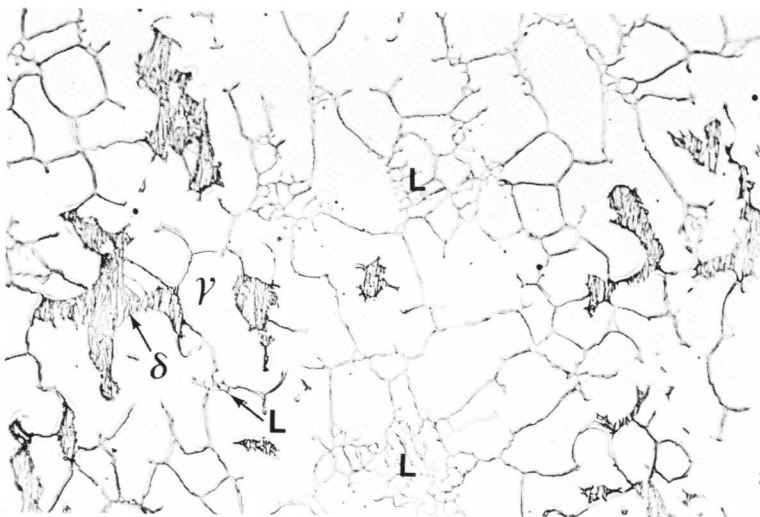


Figure 7

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1415^{\circ}\text{C}$

γ growing into both δ -dendrites and liquid.

100 μm $\times 150$

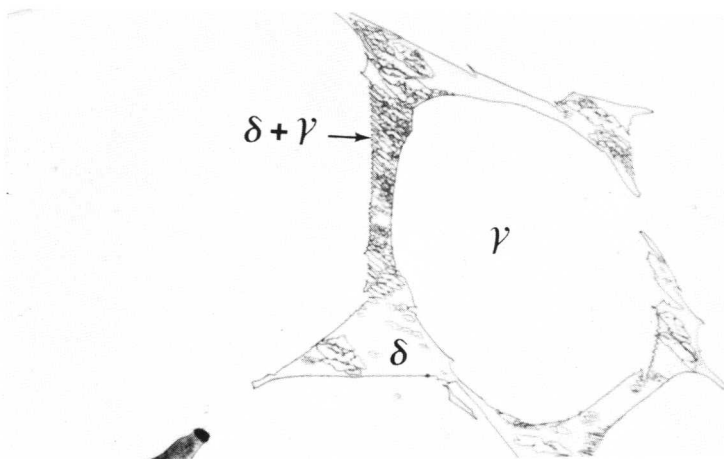


Figure 8

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1320^{\circ}\text{C}$

Austenite precipitated in interdendritic ferrite during quenching. Dark structure in figure 10.

Figure 9

R = 2,0°C/s
 Tq = 1320°C
 5,5 vol-% ferrite, dendritic (D δ) and interdendritic (ID δ).

× 150 100 μ m

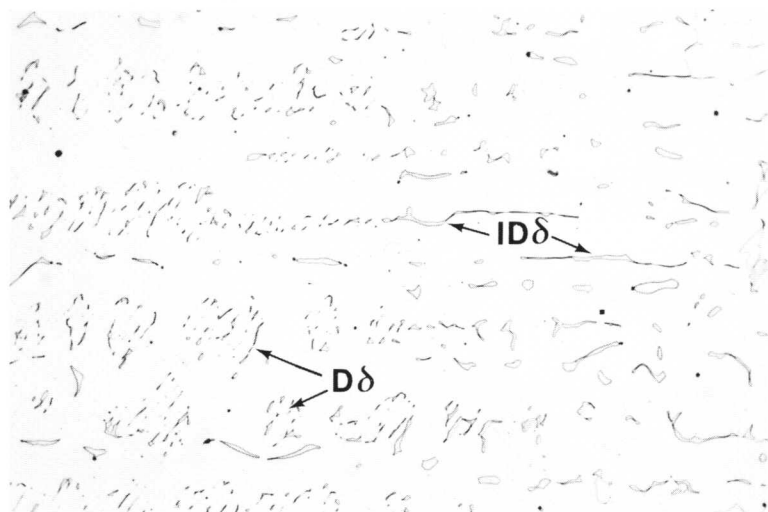


Figure 10

R = 0,5°C/s
 Tq = 1320°C
 5,6 vol-% ferrite, mainly interdendritic (ID δ).

× 150 100 μ m

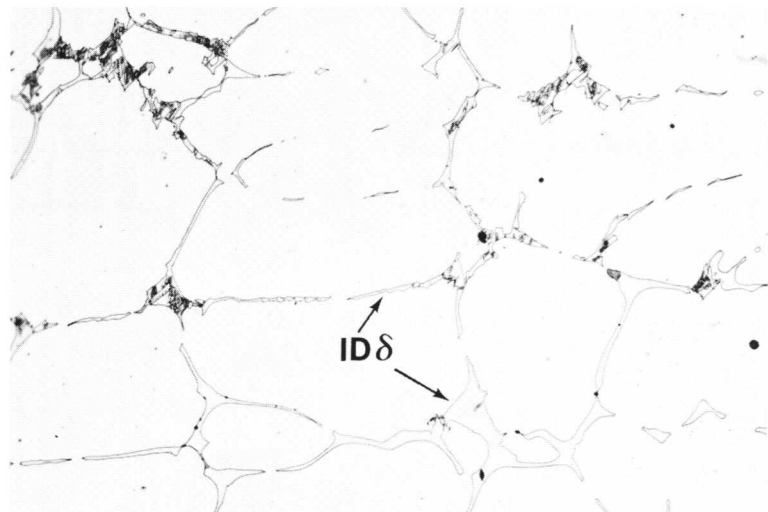


Figure 11

R = 0,5°C/s
 Tq = 1200°C
 3,5 vol-% ferrite, mainly interdendritic (ID δ).

× 150 100 μ m

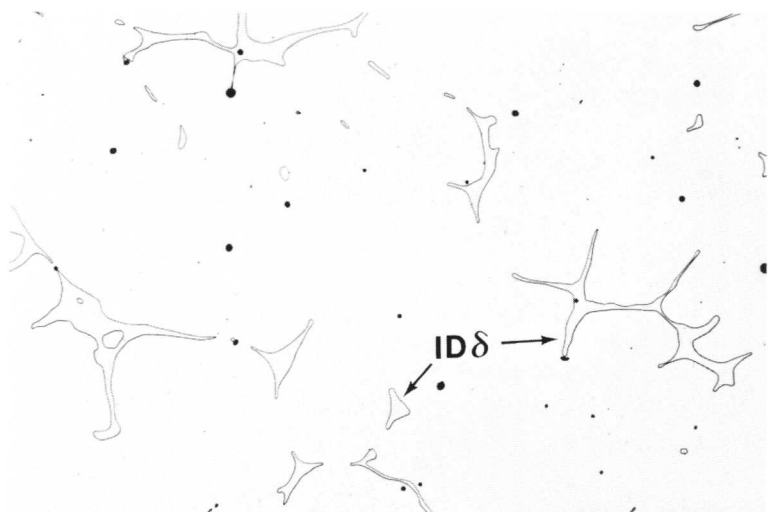
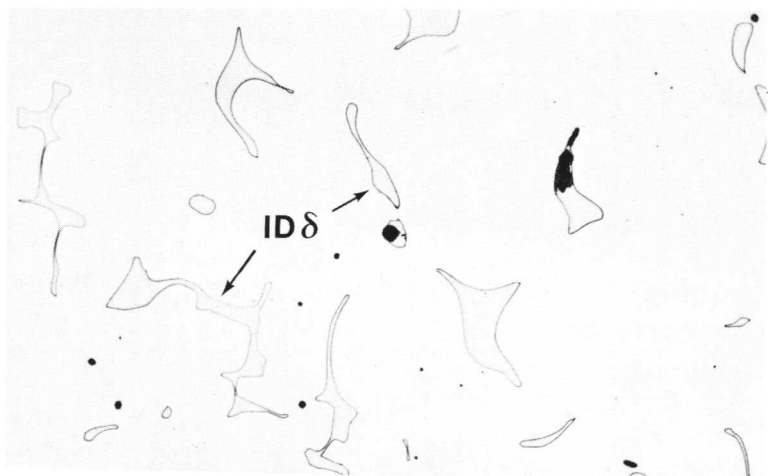


Figure 12

R = 0,1°C/s
 Tq = 1320°C
 4,4 vol-% ferrite, mainly interdendritic (ID δ).

100 μ m



STEEL 408. 0,05 % C 18 % Cr 13 % Ni 2,5 % Mo STAINLESS STEEL

Designations

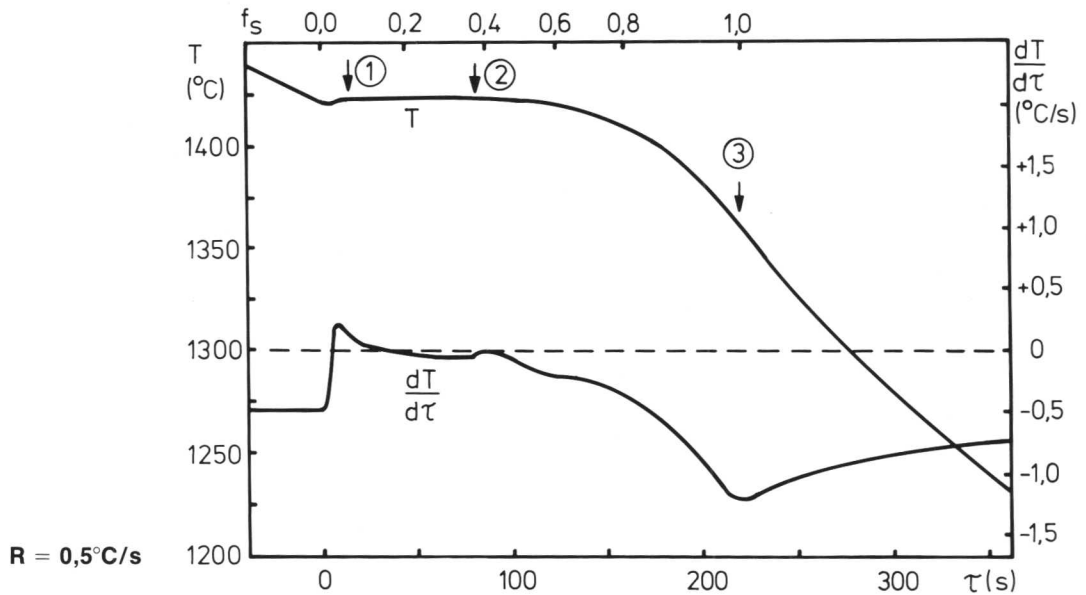
SIS	AISI	Werkstoff Nr
2343	316	1.4401

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Al _{tot}	N
0,048	0,63	1,65	0,018	0,007	17,7	13,4	2,68	0,15	0,07	0,004	0,045

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,42$$

Thermal Analysis



	Average Cooling Rate, R, (°C/s)		
	2,0	0,5	0,1
Liquidus temperature, ferritic and austenitic primary phases, °C ①	1419	1423	1421
Temperature of maximum rate of formation of austenite, °C ②	1414	1422	1415
Solidus temperature, °C	1330	1360	1370
Solidification range, °C	85	60	50
Solidification time, s	100	220	670
Fraction solidified as ferrite, %	<36	<35	<34

Precipitates

Interdendritic ferrite, (see figures, 5, 8 – 11).

Microsegregation

Element	Mn	Cr	Ni	Mo
I	1,6	1,2	1,2	2,1
P _{ID}		1,2	0,7	

R = 0,5 °C/s
T_q = 1305 °C

Partly solidified

Figure 1

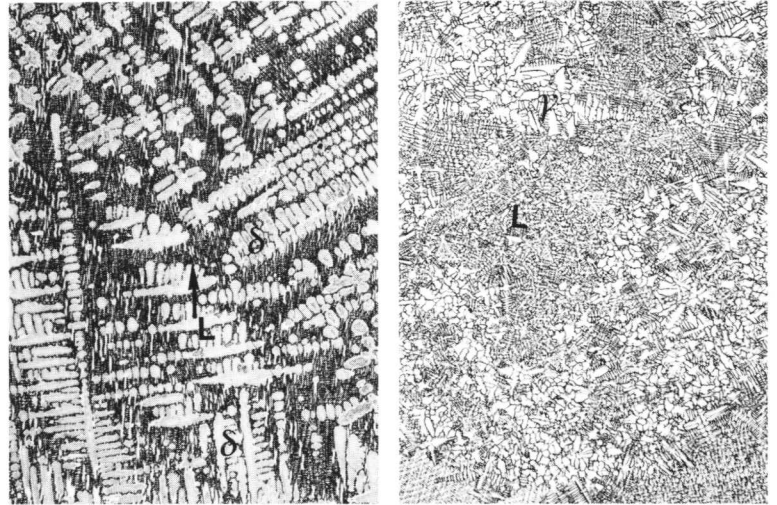
$R = 0,5^{\circ}\text{C/s}$

$T_q = 1423^{\circ}\text{C}$

$d = 50\ \mu\text{m}$

δ - and γ - dendrites, growing simultaneously, and quenched liquid (L).

$\times 25$ 



Completely solidified

Figure 2

$R = 2,0^{\circ}\text{C/s}$

$T_q = 1305^{\circ}\text{C}$

$d = 55\ \mu\text{m}$

Former δ -dendrites, γ -dendrites and interdendritic ferrite. Some dendritic ferrite can also be seen, (compare figure 8).

$\times 25$ 

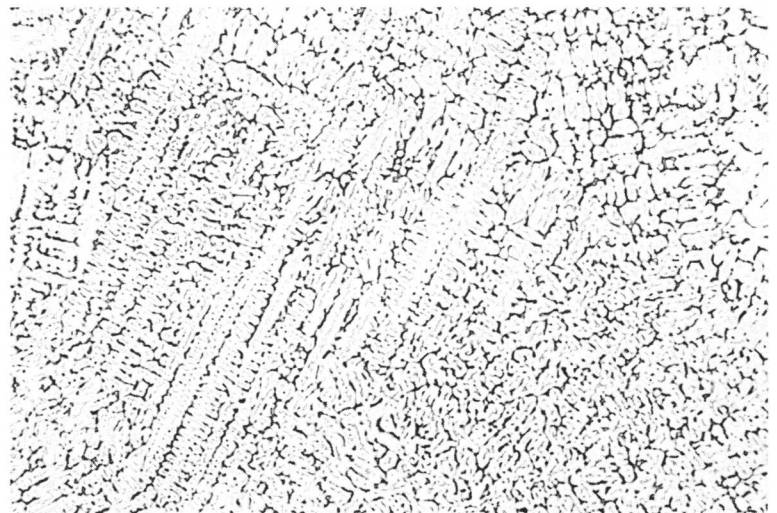


Figure 3

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1305^{\circ}\text{C}$

$d = 85\ \mu\text{m}$

Former δ -dendrites, γ -dendrites and interdendritic ferrite, (compare figure 9).

$\times 25$ 



Figure 4

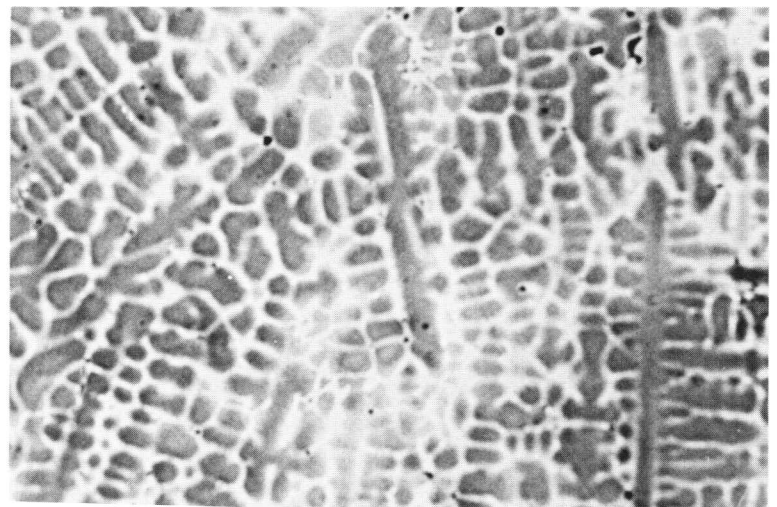
$R = 0,1^{\circ}\text{C/s}$

$T_q = 1305^{\circ}\text{C}$

$d = 140\ \mu\text{m}$

Former δ -dendrites and γ -dendrites. White interdendritic areas, (compare figure 11).

$\times 25$ 



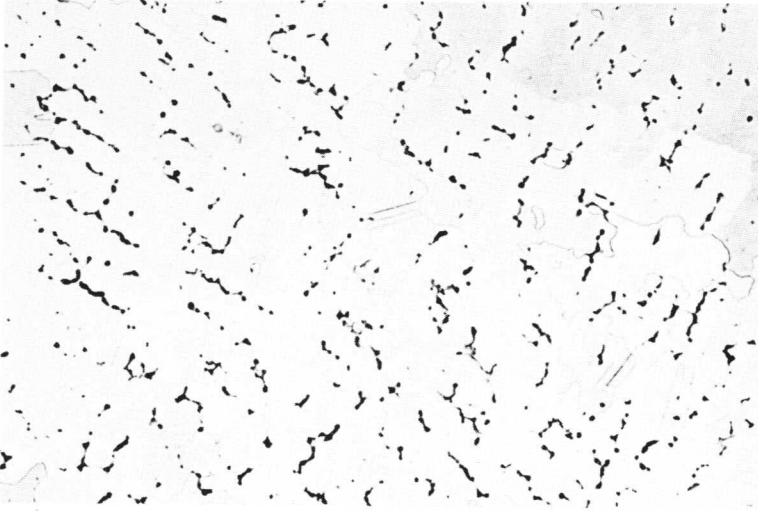


Figure 5

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1200^{\circ}\text{C}$

($d_{1200} = 190 \mu\text{m}$)

Former δ -dendrites, γ -dendrites and interdendritic ferrite, (compare figure 10).

$400 \mu\text{m} \quad \times 25$

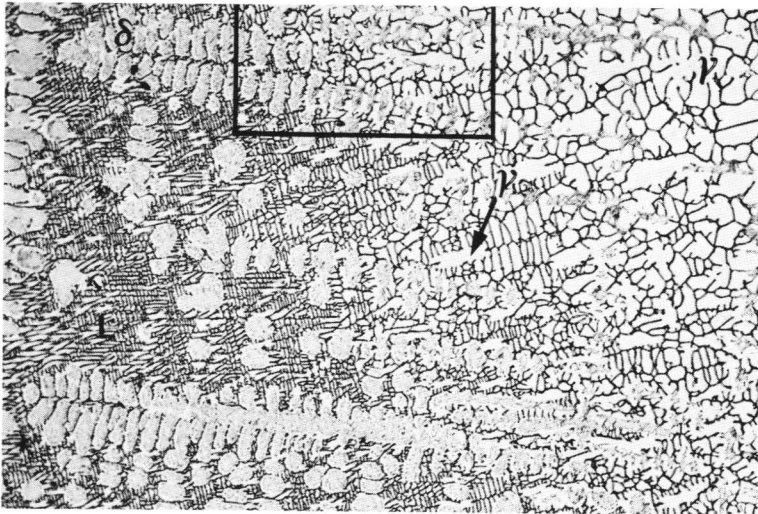


Figure 6

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1420^{\circ}\text{C}$

Simultaneous growth of δ - and γ - dendrites. The transformation of δ into γ can also be seen, (peritectic reaction and transformation, compare figure 7).

L = quenched liquid.

$200 \mu\text{m} \quad \times 50$

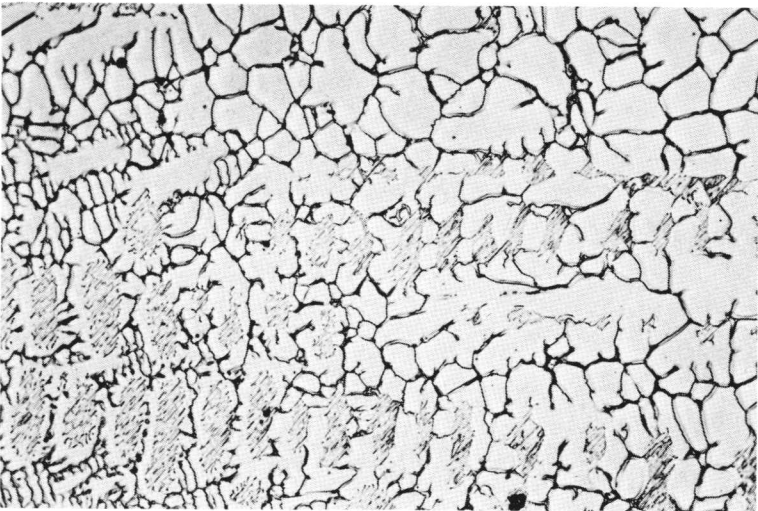


Figure 7

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1420^{\circ}\text{C}$

Transformation of δ -dendrites.

Detail of figure 6.

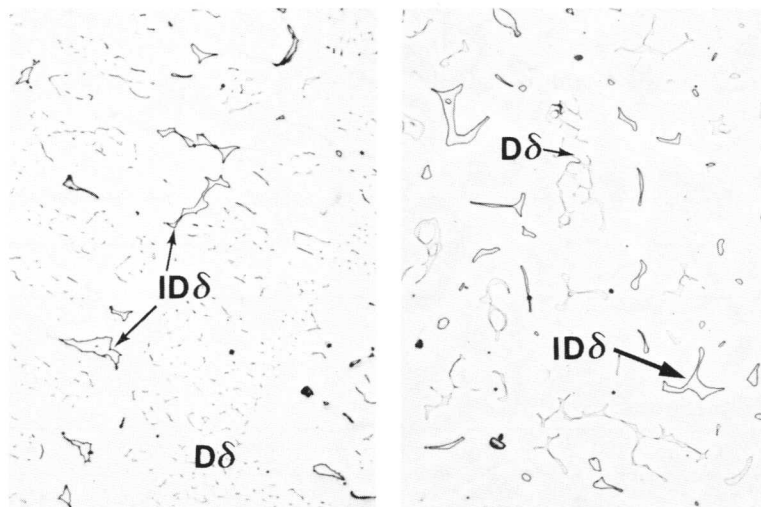
$100 \mu\text{m} \quad \times 150$

Figure 8

R = 2,0°C/s

T_q = 1305°C

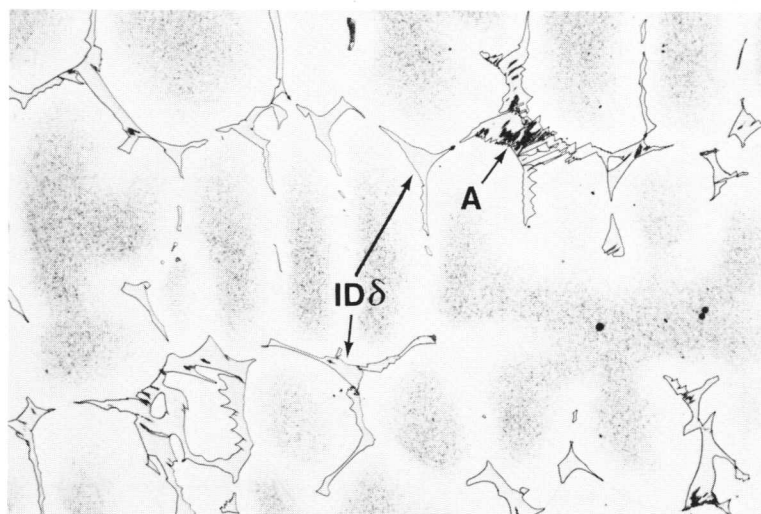
4,0 vol-% ferrite, dendritic (Dδ) and interdendritic (IDδ).

× 150 **Figure 9**

R = 0,5°C/s

T_q = 1305°C

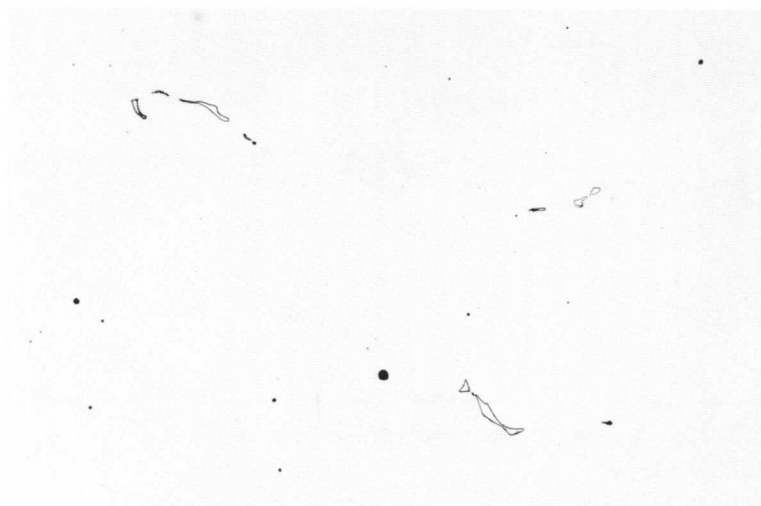
5,0 vol-% ferrite, mainly interdendritic (IDδ). (Dark structure (A), austenite precipitated in ferrite during quenching.)

× 150 **Figure 10**

R = 0,5°C/s

T_q = 1200°C

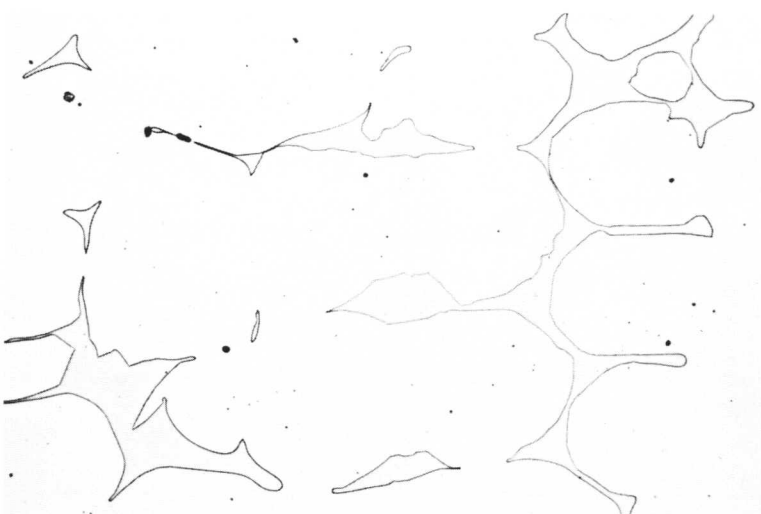
0,1 vol-% ferrite, mainly interdendritic.

× 150 **Figure 11**

R = 0,1°C/s

T_q = 1305°C

5,5 vol-% ferrite, mainly interdendritic.

× 150 

STEEL 409. 0,02 % C 17 % Cr 13 % Ni 2,5 % Mo 0,2 % N STAINLESS STEEL

Designations

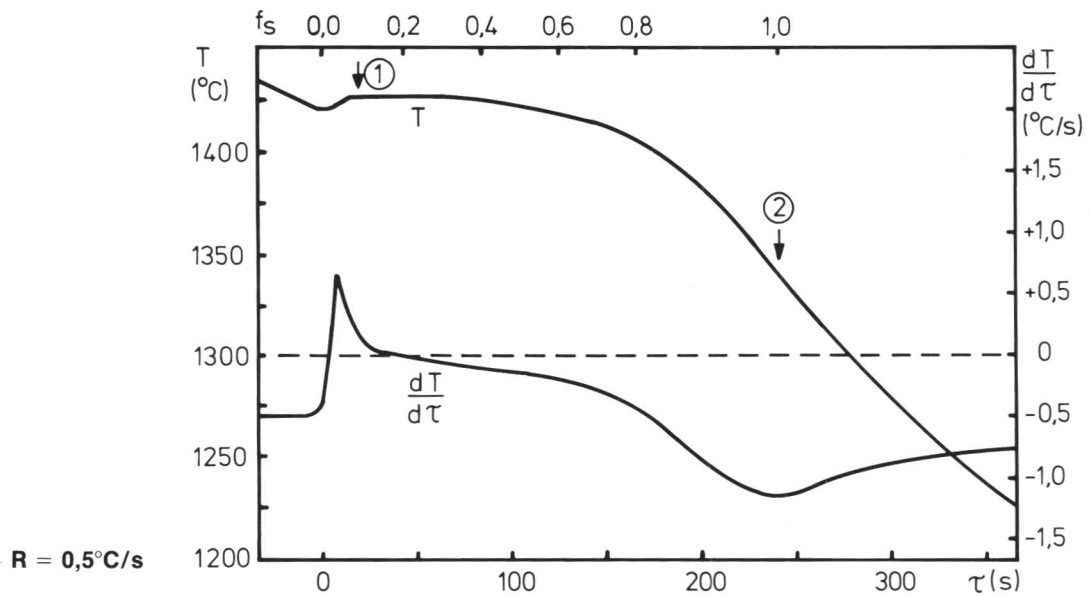
SIS	AISI	Werkstoff Nr
2375	316 N	1.4429

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Al _{tot}	N
0,024	0,58	1,79	0,009	0,011	17,4	12,8	2,77	0,03	0,03	0,002	0,20

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,32$$

Thermal Analysis



	Average Cooling Rate, R, (°C/s)		
	2,0	0,5	0,1
Liquidus temperature, austenitic primary phase, °C (1)	1411	1421	1422
Solidus temperature, °C (2)	1310	1350	1370
Solidification range, °C	105	70	50
Solidification time, s	85	245	640

Precipitates

Interdendritic ferrite, (see figures 5, 8 – 11).

Microsegregation

Element	Mn	Cr	Ni	Mo
I	1,6	1,2	1,1	2,1

$R = 0,5^\circ\text{C/s}$
 $T_q = 1305^\circ\text{C}$

Partly solidified


Figure 1

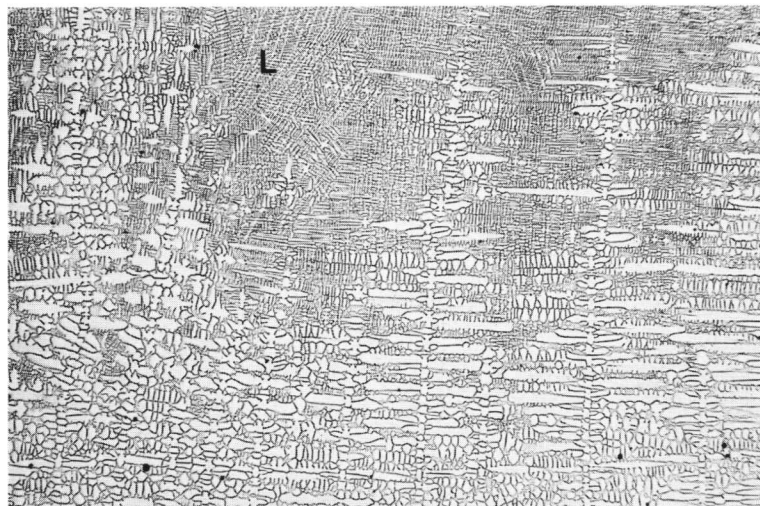
$R = 0,5^{\circ}\text{C/s}$

$T_q = 1415^{\circ}\text{C}$

$d = 40\ \mu\text{m}$

γ -dendrites and quenched liquid (L).

$\times 25$ 



Completely solidified

Figure 2

$R = 2,0^{\circ}\text{C/s}$

$T_q = 1305^{\circ}\text{C}$

$d = 45\ \mu\text{m}$

Figures 2–4: γ -dendrites and interdendritic ferrite.


$\times 25$ 



Figure 3

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1305^{\circ}\text{C}$

$d = 70\ \mu\text{m}$

$\times 25$ 




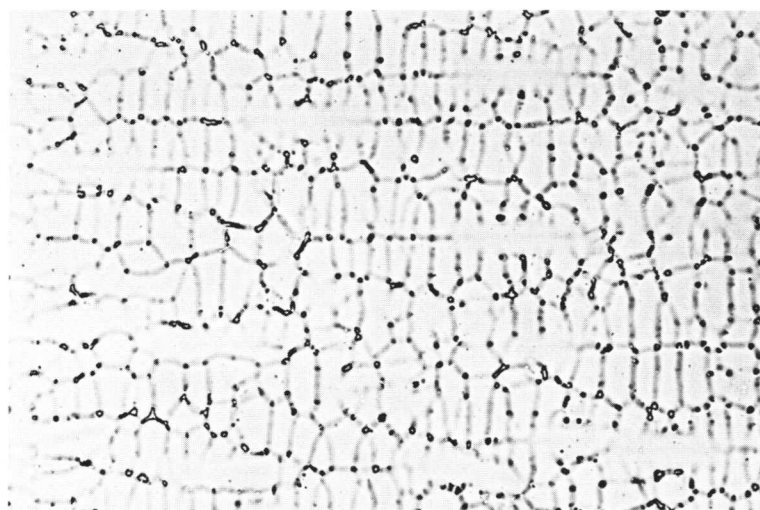
Figure 4

$R = 0,1^{\circ}\text{C/s}$

$T_q = 1305^{\circ}\text{C}$

$d = 105\ \mu\text{m}$

$\times 25$ 



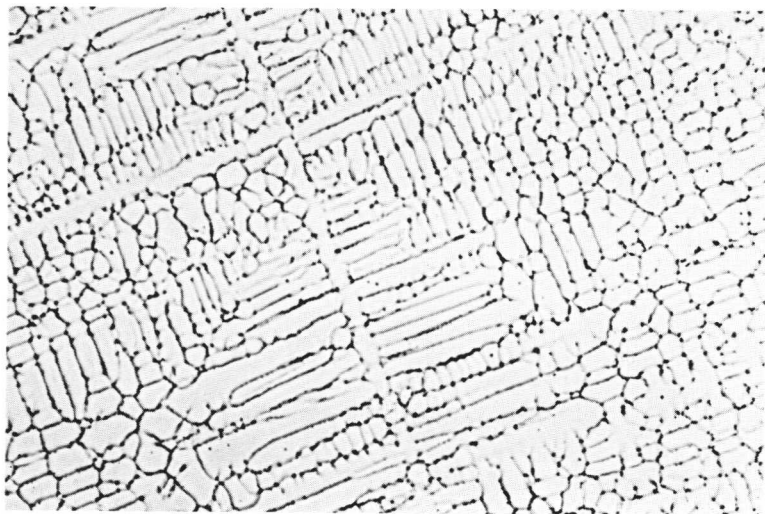


Figure 5

$R = 0,5^{\circ}\text{C/s}$
 $T_q = 1200^{\circ}\text{C}$
 $(d_{1200} = 75 \mu\text{m})$
 γ -dendrites and interdendritic ferrite.

$400 \mu\text{m} \times 25$

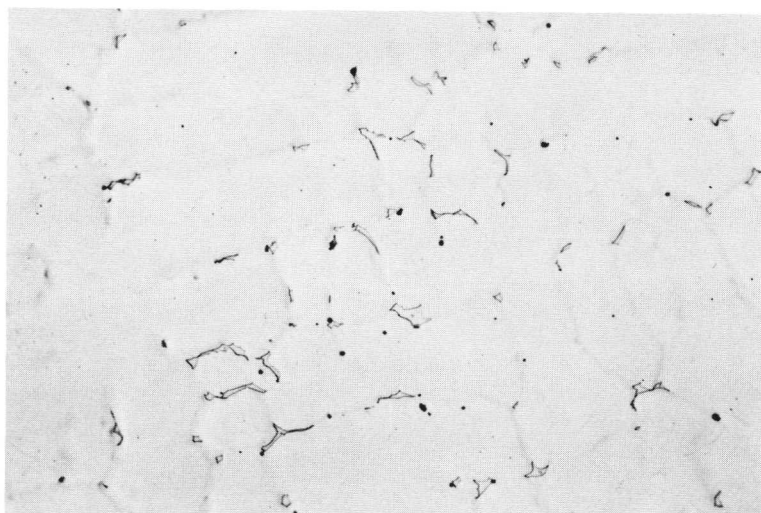


Figure 6

$R = 2,0^{\circ}\text{C/s}$
 $T_q = 1305^{\circ}\text{C}$
 0,8 vol-% interdendritic ferrite.

$100 \mu\text{m} \times 150$

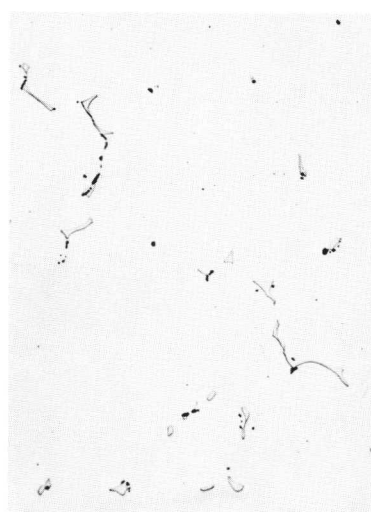
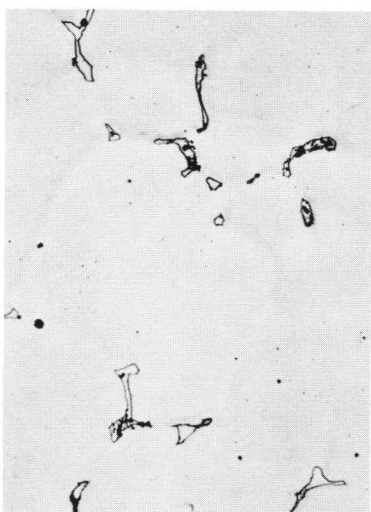


Figure 7

$R = 0,5^{\circ}\text{C/s}$
 $T_q = 1305^{\circ}\text{C}$
 0,8 vol-% inter-
 dendritic ferrite.

$100 \mu\text{m} \times 150$

Figure 8

$R = 0,5^{\circ}\text{C/s}$
 $T_q = 1200^{\circ}\text{C}$
 0,1 vol-% inter-
 dendritic ferrite.

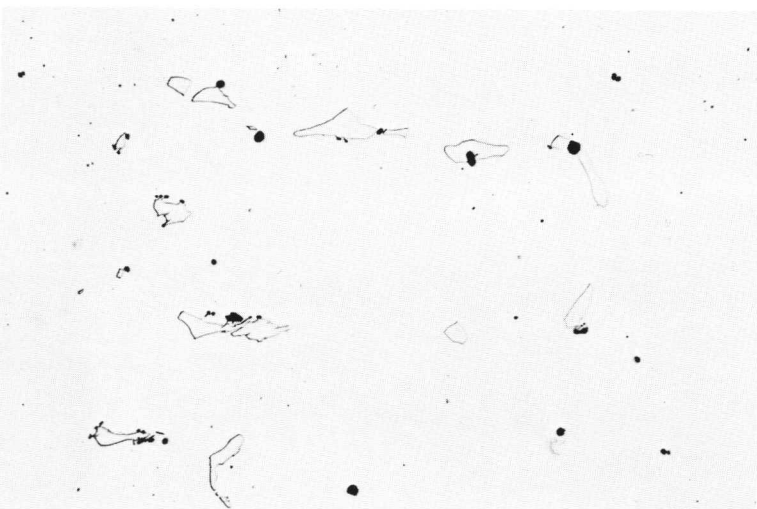


Figure 9

$R = 0,1^{\circ}\text{C/s}$
 $T_q = 1305^{\circ}\text{C}$
 0,8 vol-% interdendritic ferrite.

$100 \mu\text{m} \times 150$

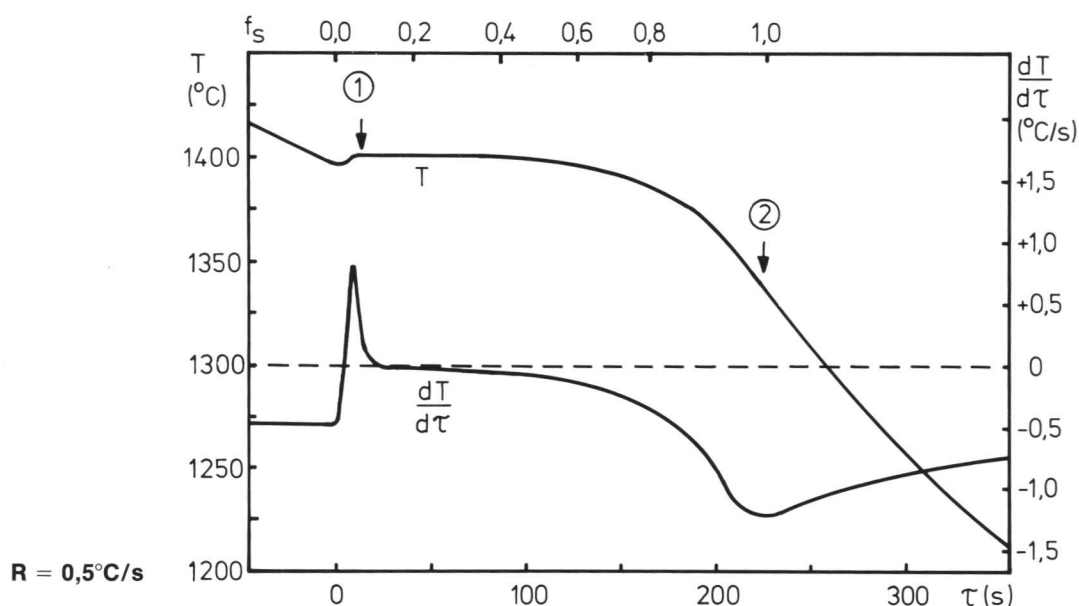
STEEL 410. 0,01 % C 25 % Cr 22 % Ni 2 % Mo STAINLESS STEEL**Designations**

SIS	AISI	Werkstoff Nr
—	—	—

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Ti	Al _{tot}	N
0,008	0,24	1,77	0,009	0,008	25,1	22,2	2,3	0,02	0,02	0,08	0,002	0,067

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,21$$

Thermal Analysis

	Average Cooling Rate, R, ($^\circ\text{C/s}$)		
	2,0	0,5	0,1
Liquidus temperature, austenitic primary phase, $^\circ\text{C}$ ①	1401	1402	1401
Solidus temperature, $^\circ\text{C}$ ②	1335	1345	1355
Solidification range, $^\circ\text{C}$	65	60	45
Solidification time, s	95	225	700

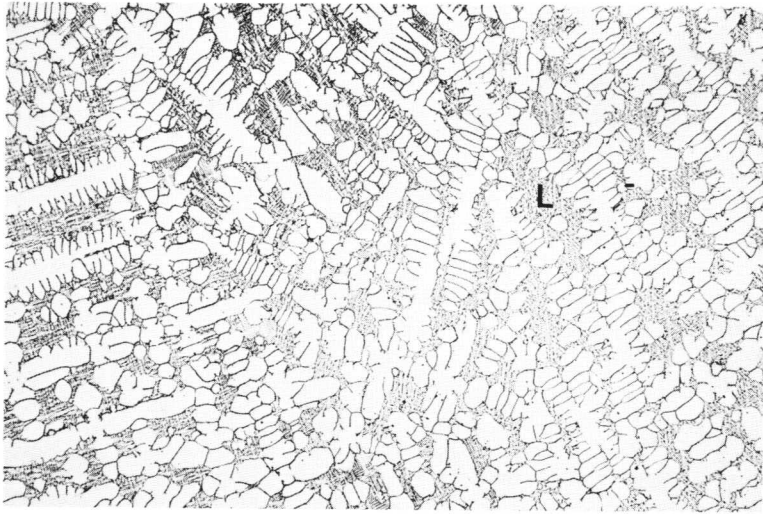
Precipitates

1. Interdendritic ferrite, (see figures 8 – 11).
2. Sigma-phase, (see figures 6 and 7).

Microsegregation

Element	Mn	Cr	Ni	Mo
I	1,6	1,2	1,1	2,3
P _{ID}		1,2	0,8	

$R = 0,5^\circ\text{C/s}$
 $T_q = 1310^\circ\text{C}$



Partly solidified

Figure 1

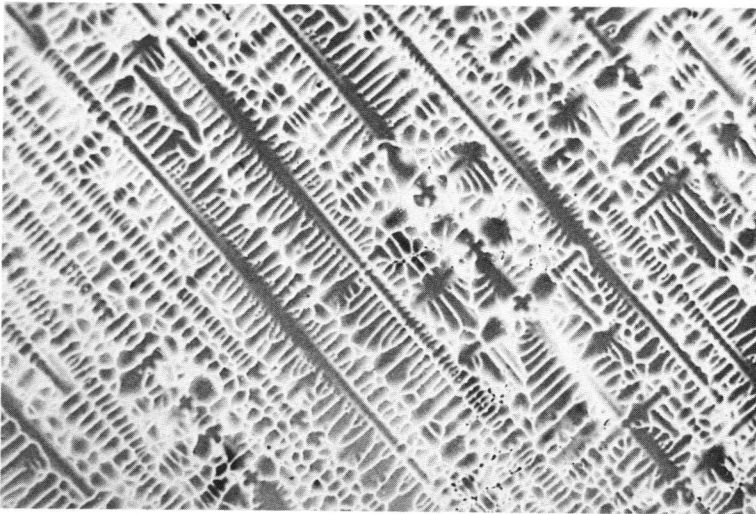
$R = 0,5^{\circ}\text{C/s}$

$T_q = 1398^{\circ}\text{C}$

$d = 60\ \mu\text{m}$

γ -dendrites and quenched liquid (L).

400 μm $\times 25$



Completely solidified

Figure 2

$R = 2,0^{\circ}\text{C/s}$

$T_q = 1310^{\circ}\text{C}$

$d = 60\ \mu\text{m}$

Figures 2–4: γ -dendrites.

White interdendritic areas.

400 μm $\times 25$

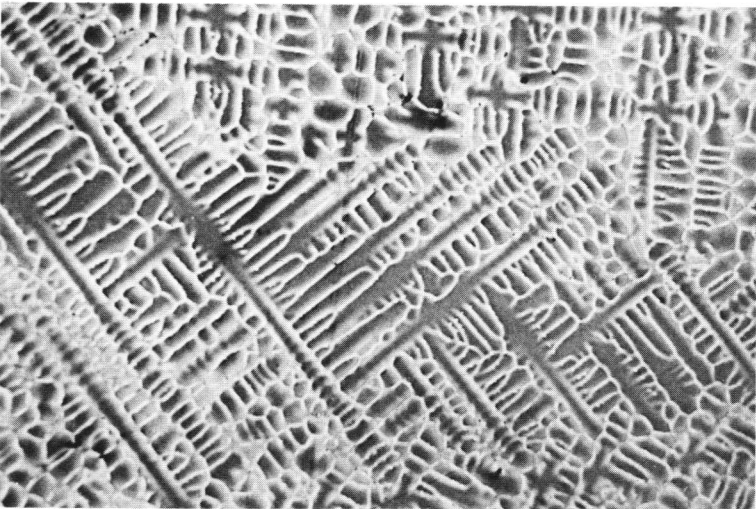


Figure 3

$R = 0,5^{\circ}\text{C/s}$

$T_q = 1310^{\circ}\text{C}$

$d = 80\ \mu\text{m}$

400 μm $\times 25$

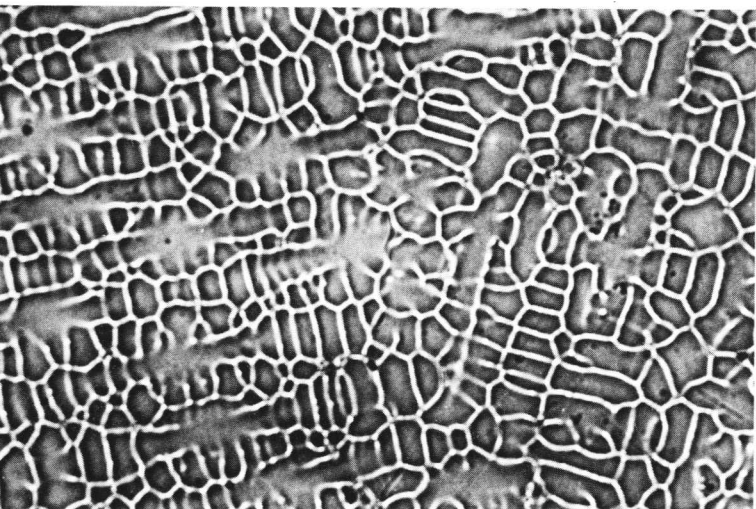


Figure 4

$R = 0,1^{\circ}\text{C/s}$

$T_q = 1310^{\circ}\text{C}$

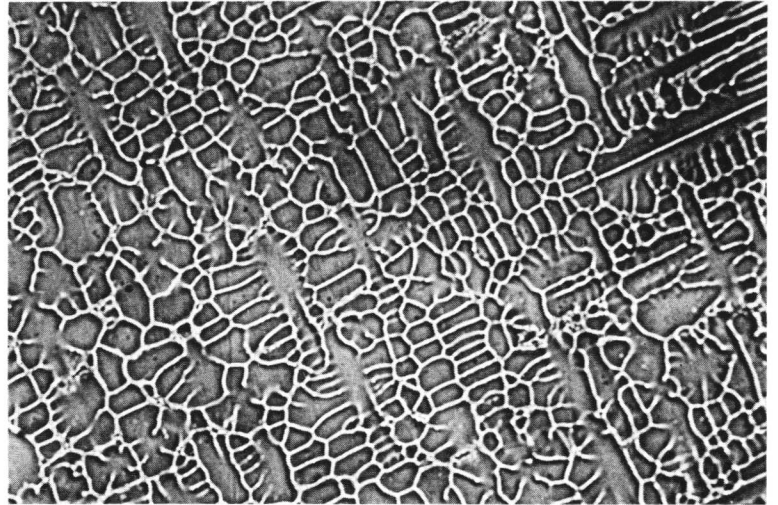
$d = 160\ \mu\text{m}$

400 μm $\times 25$

Figure 5

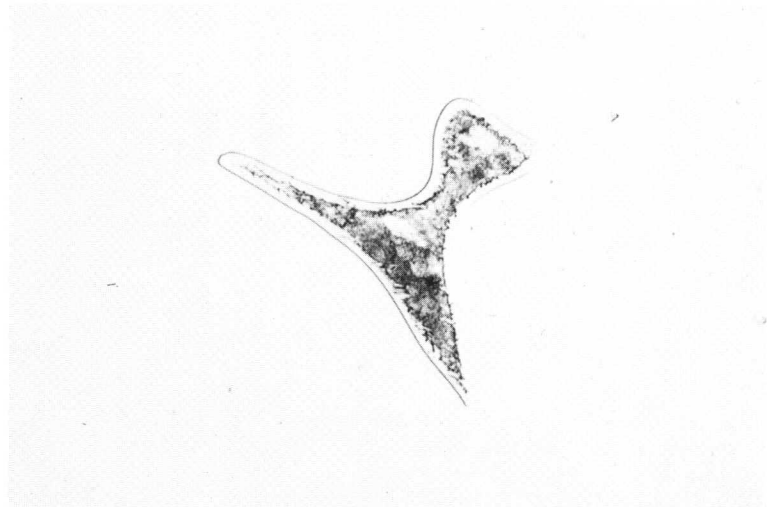
R = 0,5°C/s
 T_q = 1200°C
 (d₁₂₀₀ = 110 μm)
 γ-dendrites.
 White interdendritic areas.

× 25 400 μm

**Figure 6**

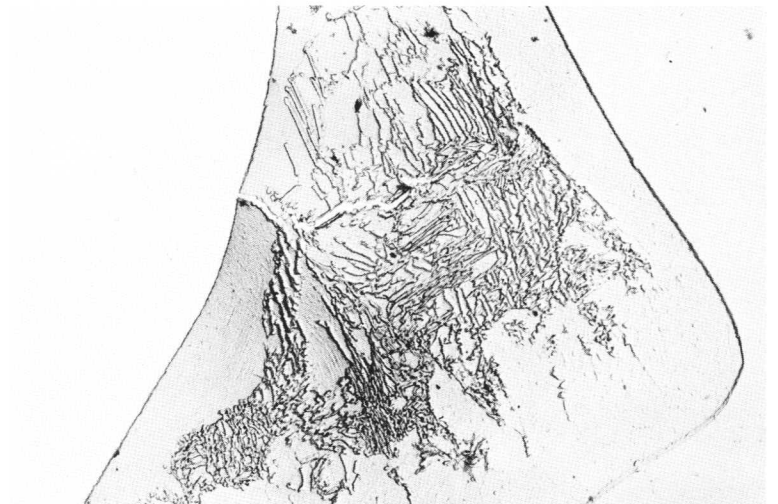
R = 0,1°C/s
 T_q = 1310°C
 Sigma-phase precipitated in ferrite, (dark structure in figures 9 and 11).

× 600 25 μm

**Figure 7**

R = 0,1°C/s
 T_q = 1310°C
 Sigma-phase precipitated in ferrite.
 (Electron micrograph.)

× 3 000 5 μm



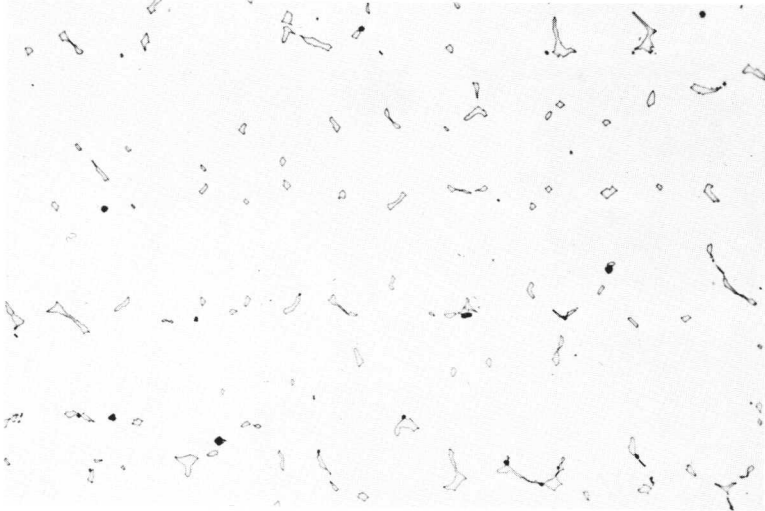


Figure 8

R = 2,0°C/s

T_q = 1310°C

1,0 vol-% interdendritic ferrite.

100 μm × 150

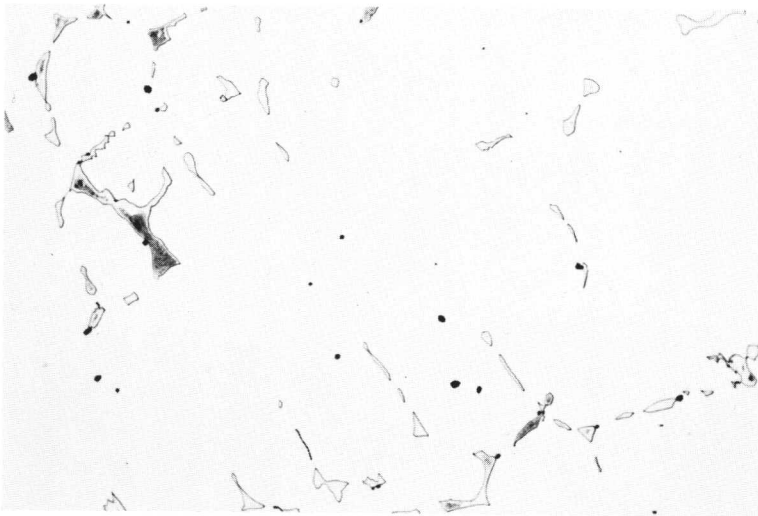


Figure 9

R = 0,5°C/s

T_q = 1310°C

1,0 vol-% interdendritic ferrite.

100 μm × 150



Figure 10

R = 0,5°C/s

T_q = 1200°C

0,7 vol-% interdendritic ferrite.

100 μm × 150

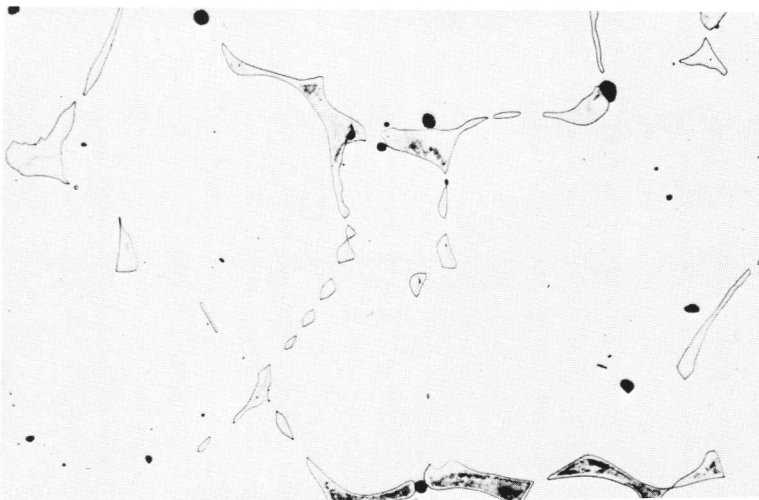


Figure 11

R = 0,1°C/s

T_q = 1310°C

1,0 vol-% interdendritic ferrite.

100 μm × 150

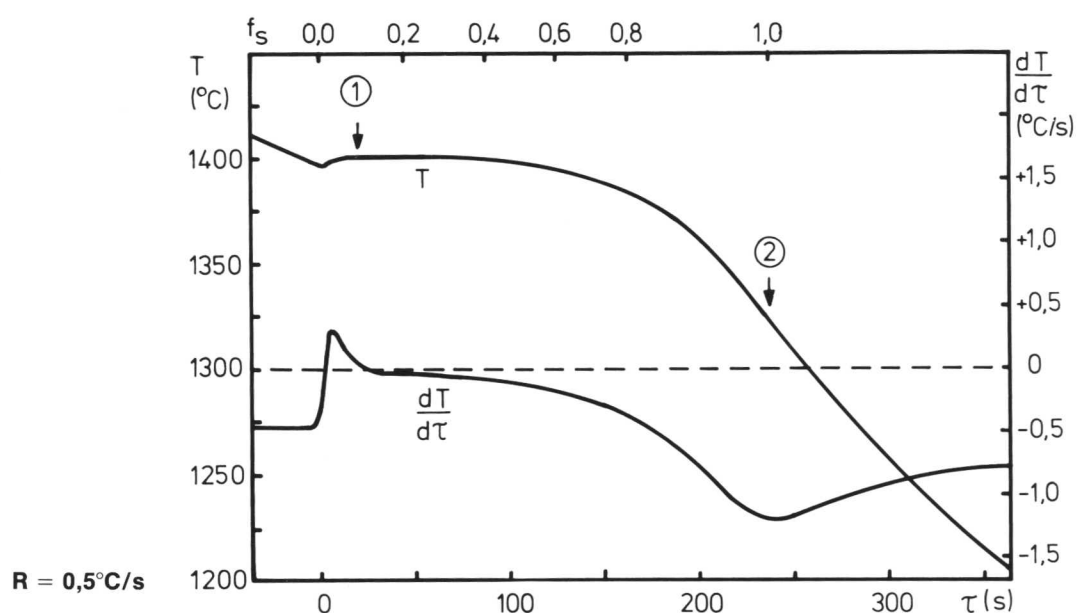
STEEL 411. 0,07 % C 24 % Cr 20 % Ni HEAT RESISTANT STEEL**Designations**

SIS	AISI	Werkstoff Nr
2361	310 S	1.4842

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Ti	Al _{tot}	N
0,055	1,20	1,75	0,011	0,008	24,2	20,4	0,08	0,02	0,03	0,09	0,015	0,051

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,15$$

Thermal Analysis

	Average Cooling Rate, R , ($^\circ\text{C/s}$)		
	2,0	0,5	0,1
Liquidus temperature, austenitic primary phase, $^\circ\text{C}$ (1)	1399	1401	1399
Solidus temperature, $^\circ\text{C}$ (2)	1315	1330	1350
Solidification range, $^\circ\text{C}$	80	70	50
Solidification time, s	105	230	750

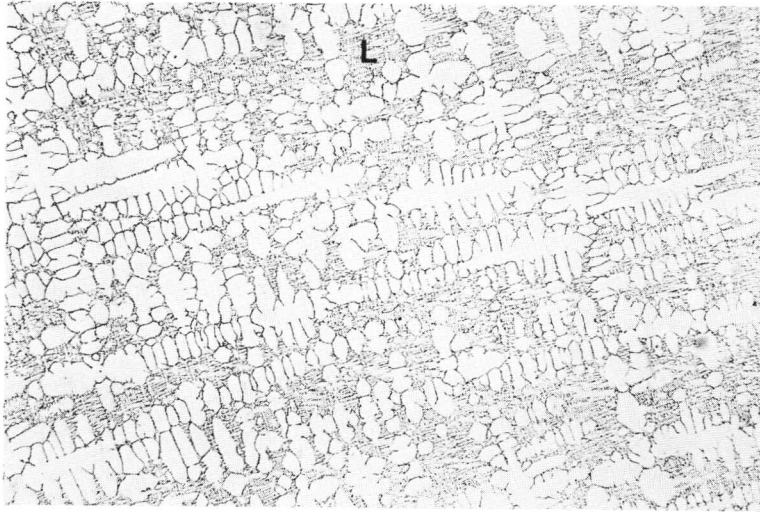
Precipitates

Interdendritic ferrite, (see figures 6–8).

Microsegregation

Element	Si	Mn	Cr	Ni
I	2,4	1,9	1,2	1,2

$R = 0,5^\circ\text{C/s}$
 $T_q = 1290^\circ\text{C}$

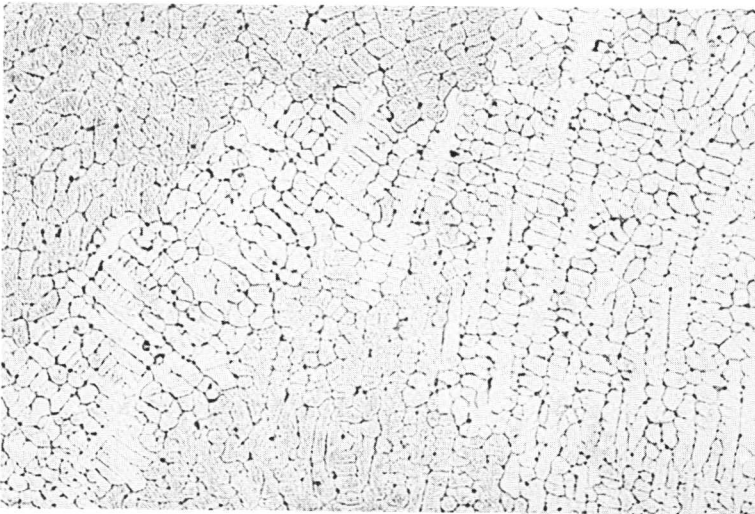


Partly solidified

Figure 1

R = 0,5°C/s
 T_q = 1395°C
 d = 65 μm
 γ-dendrites and quenched liquid (L).

400 μm × 25



Completely solidified

Figure 2

R = 2,0°C/s
 T_q = 1290°C
 d = 55 μm
 Figures 2–4: γ-dendrites.

400 μm × 25

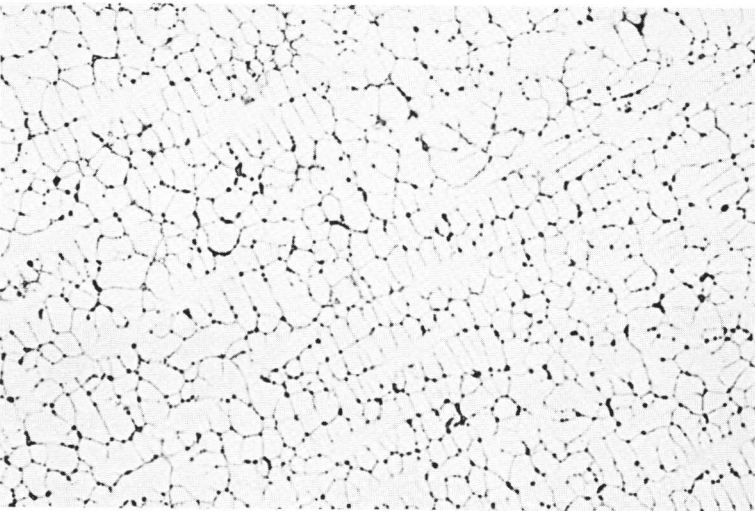


Figure 3

R = 0,5°C/s
 T_q = 1290°C
 d = 85 μm

400 μm × 25

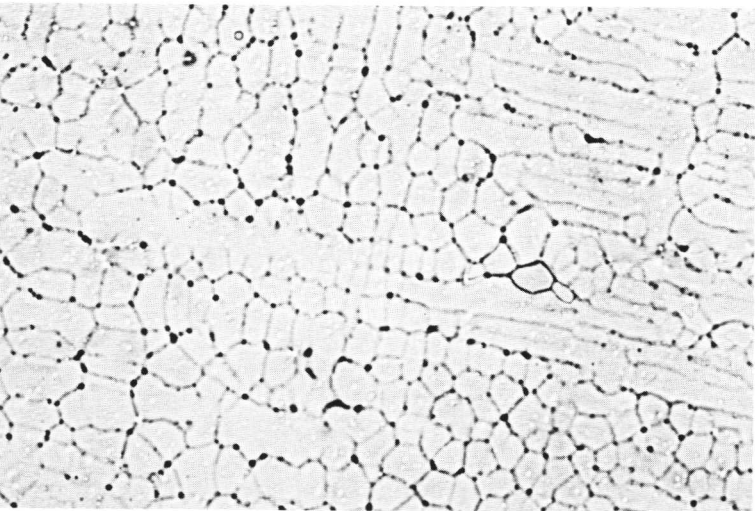


Figure 4

R = 0,1°C/s
 T_q = 1290°C
 d = 125 μm

400 μm × 25

Figure 5

R = 0,5°C/s
 Tq = 1200°C
 (d₁₂₀₀ = 90 μm)
 γ-dendrites.

× 25 400 μm

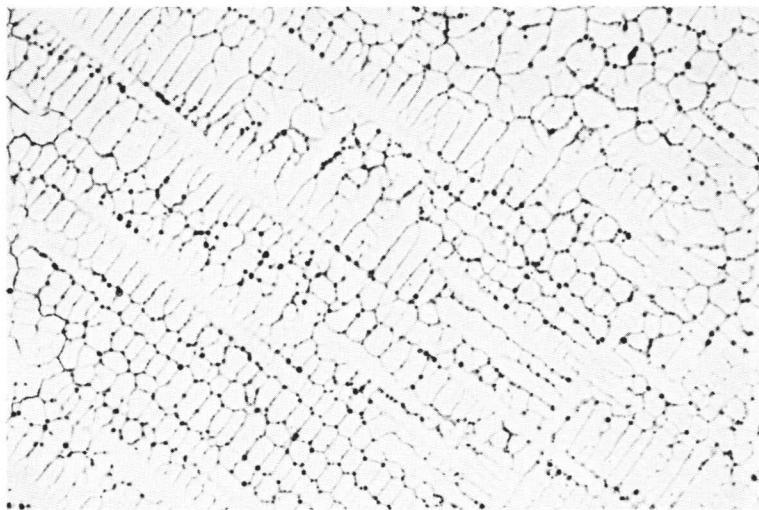


Figure 6

R = 2,0°C/s
 Tq = 1290°C
 Figures 6–8: Small amounts of interdendritic ferrite.

× 150 100 μm

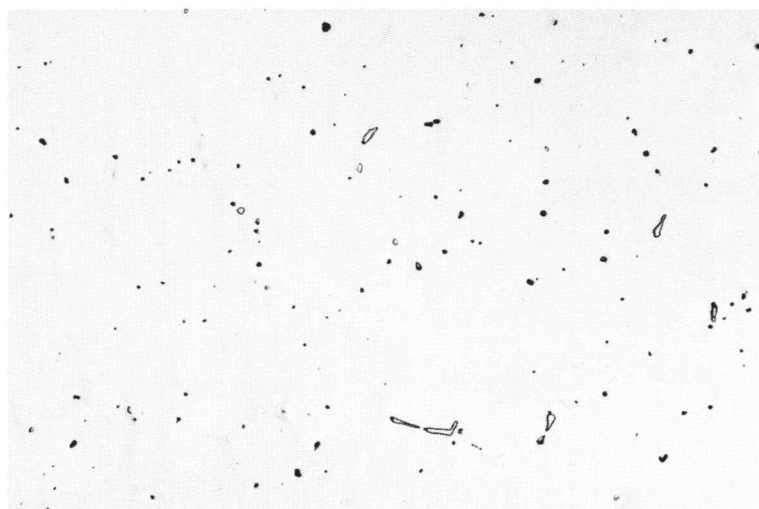


Figure 7

R = 0,5°C/s
 Tq = 1200°C

× 150 100 μm

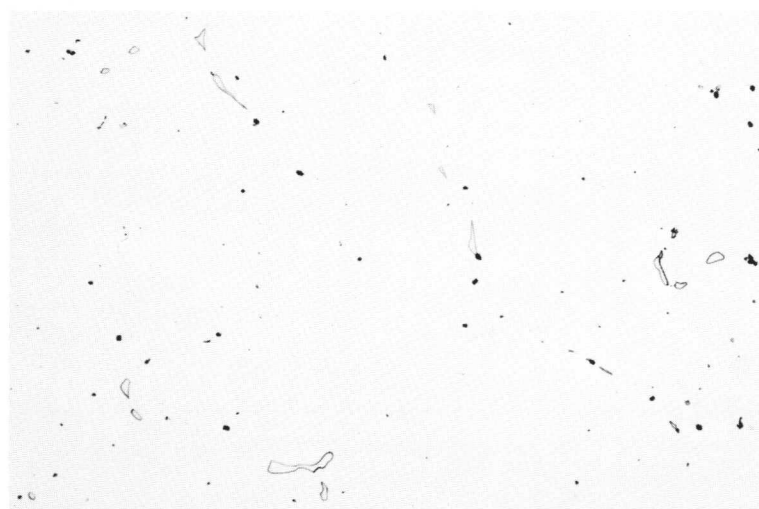
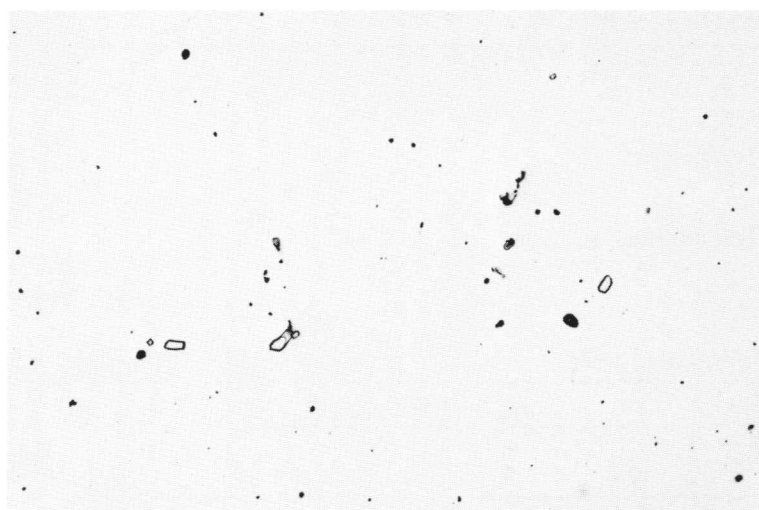


Figure 8

R = 0,1°C/s
 Tq = 1290°C

× 150 100 μm



STEEL 412. 0,1 % C 24 % Cr 20 % Ni HEAT RESISTANT STEEL

Designations

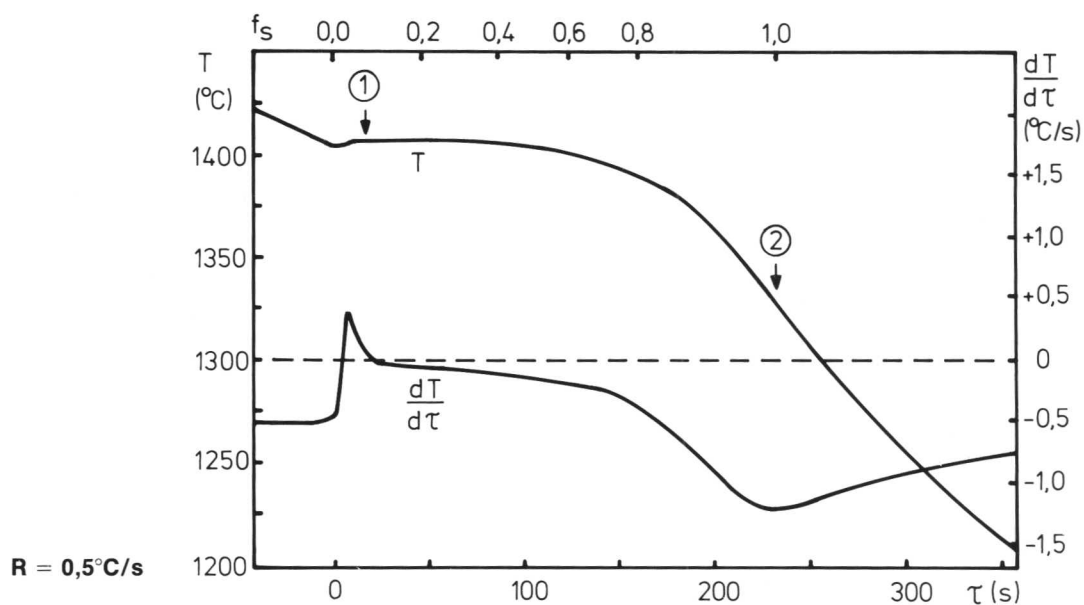
SIS	AISI	Werkstoff Nr
—	310	1.4845

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Ti	Al _{tot}	N
0,13	0,52	1,67	0,009	0,003	24,3	20,5	0,11	0,03	0,04	0,08	0,023	0,053

$$\frac{Cr_{eq}}{Ni_{eq}} = 1,03$$

Thermal Analysis



	Average Cooling Rate, R, (°C/s)		
	2,0	0,5	0,1
Liquidus temperature, austenitic primary phase, °C (1)	1405	1407	1405
Solidus temperature, °C (2)	1325	1335	1355
Solidification range, °C	80	70	50
Solidification time, s	95	230	690

Precipitates

Interdendritic ferrite, (see figures 6 – 8).

Microsegregation

Element	Si	Mn	Cr	Ni
I	2,5	1,9	1,2	1,2

$R = 0,5^\circ\text{C/s}$
 $T_q = 1300^\circ\text{C}$

Partly solidified**Figure 1**

R = 0,5°C/s

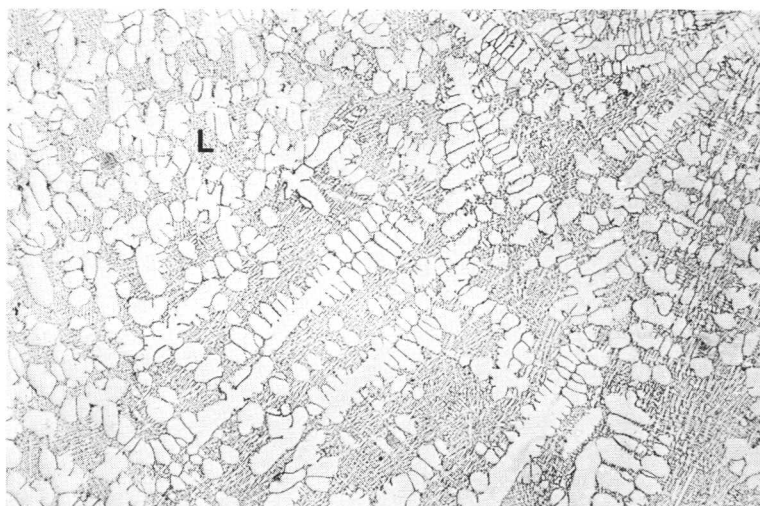
T_q = 1400°C

d = 60 μm

γ-dendrites and quenched liquid (L).

× 25

400 μm

**Completely solidified****Figure 2**

R = 2,0°C/s

T_q = 1300°C

d = 65 μm

Figures 2–4: γ-dendrites.

× 25

400 μm

**Figure 3**

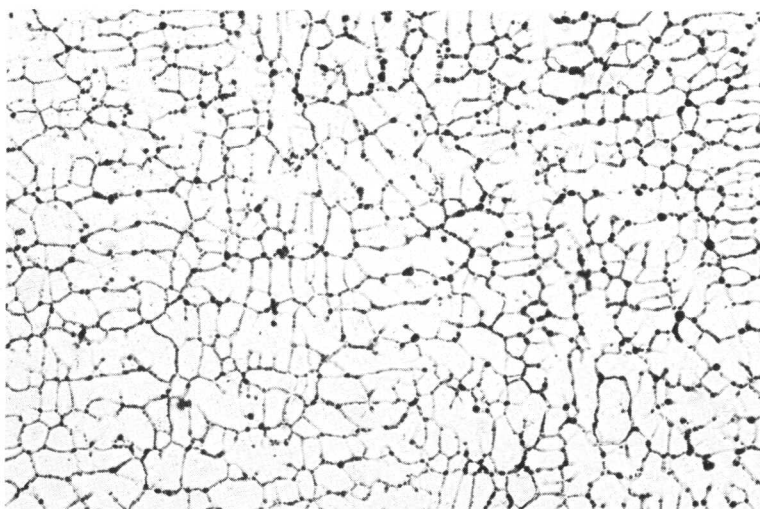
R = 0,5°C/s

T_q = 1300°C

d = 90 μm

× 25

400 μm

**Figure 4**

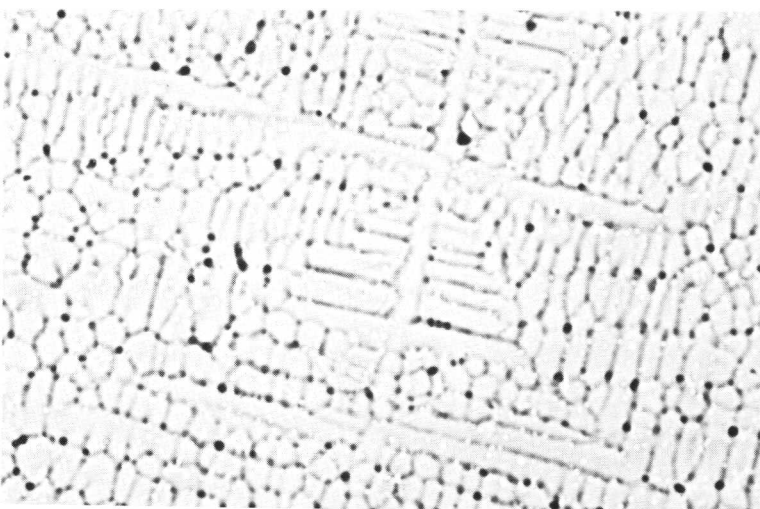
R = 0,1°C/s

T_q = 1300°C

d = 125 μm

× 25

400 μm



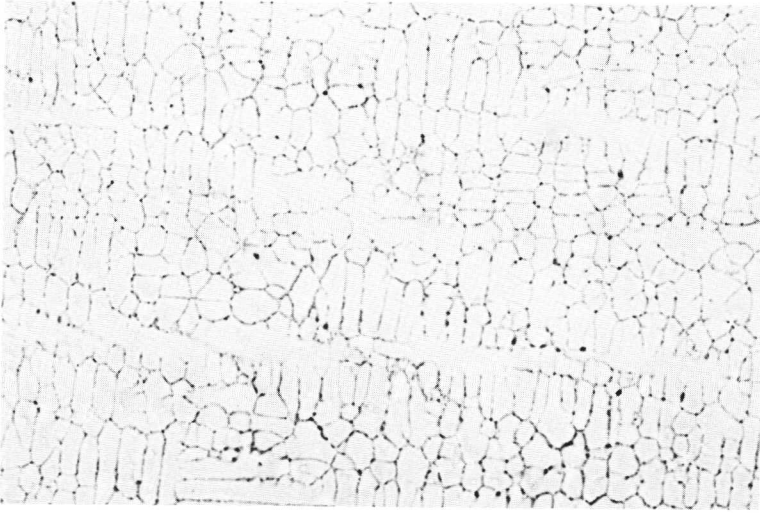


Figure 5

$R = 0,5^{\circ}\text{C/s}$
 $T_q = 1200^{\circ}\text{C}$
($d_{1200} = 100\ \mu\text{m}$)
 γ -dendrites.

400 μm $\times 25$

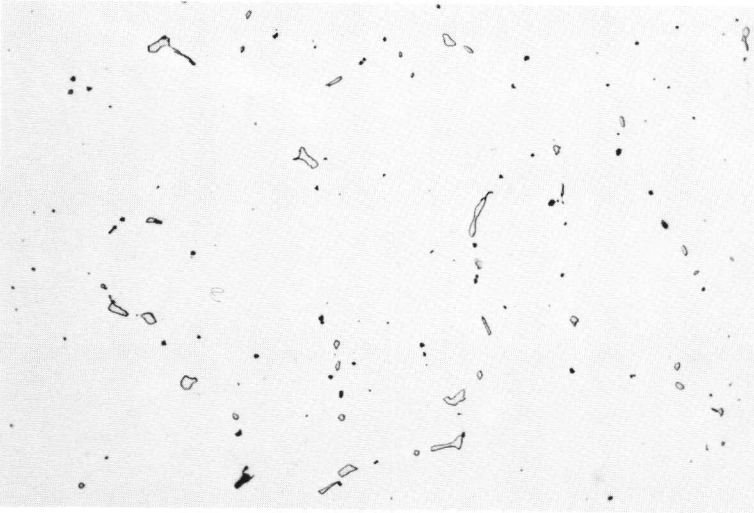


Figure 6

$R = 2,0^{\circ}\text{C/s}$
 $T_q = 1300^{\circ}\text{C}$
0,5 vol-% interdendritic ferrite.

100 μm $\times 150$

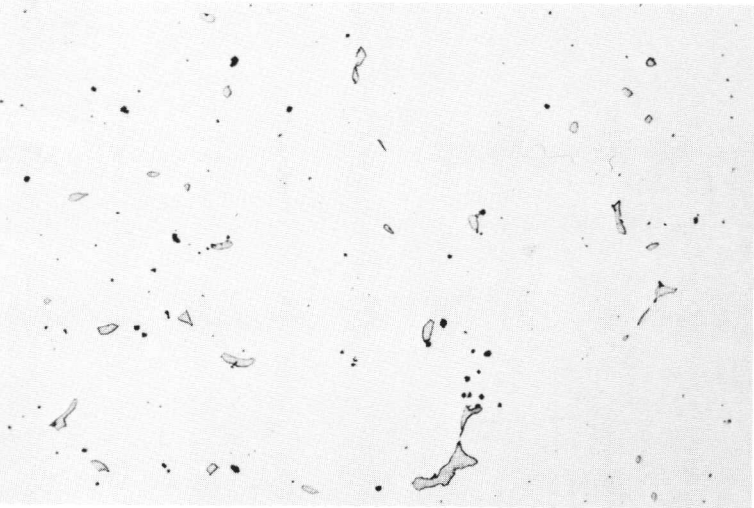


Figure 7

$R = 0,5^{\circ}\text{C/s}$
 $T_q = 1200^{\circ}\text{C}$
0,4 vol-% interdendritic ferrite.

100 μm $\times 150$

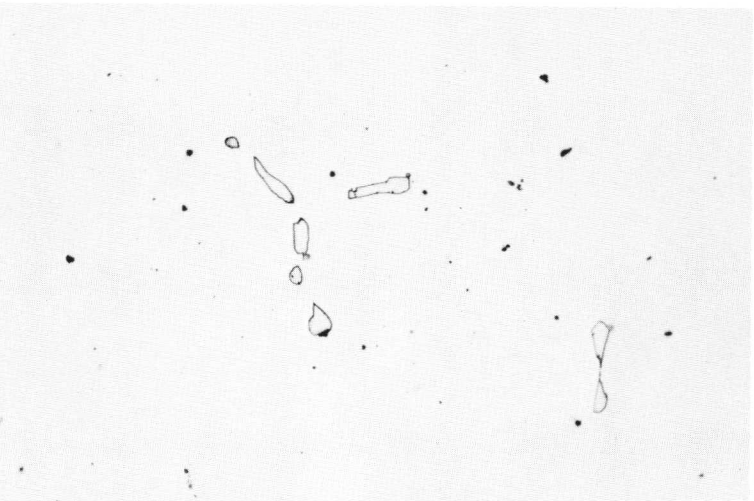


Figure 8

$R = 0,1^{\circ}\text{C/s}$
 $T_q = 1300^{\circ}\text{C}$
0,5 vol-% interdendritic ferrite.

100 μm $\times 150$

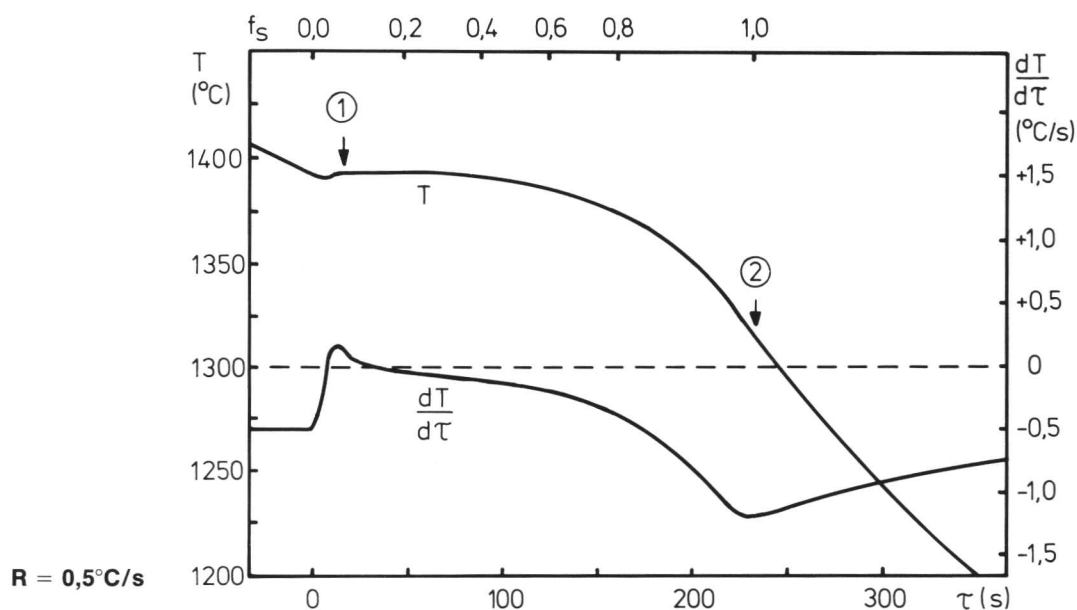
STEEL 413. 0,01 % C 19 % Cr 25 % Ni 4 % Mo 1,5 % Cu STAINLESS STEEL**Designations**

SIS	AISI	Werkstoff Nr
—	—	1.4539

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Ti	Ce	Al _{tot}	N
0,013	0,48	1,74	0,007	0,003	19,2	25,1	4,44	1,51	0,02	0,07	0,07	0,034	0,035

$$\frac{Cr_{eq}}{Ni_{eq}} = 0,94$$

Thermal Analysis**Average Cooling Rate, R, ($^\circ\text{C/s}$)**

	2,0	0,5	0,1
Liquidus temperature, austenitic primary phase, $^\circ\text{C}$ (1)	1389	1391	1391
Solidus temperature, $^\circ\text{C}$ (2)	1305	1315	1345
Solidification range, $^\circ\text{C}$	85	75	45
Solidification time, s	100	230	760

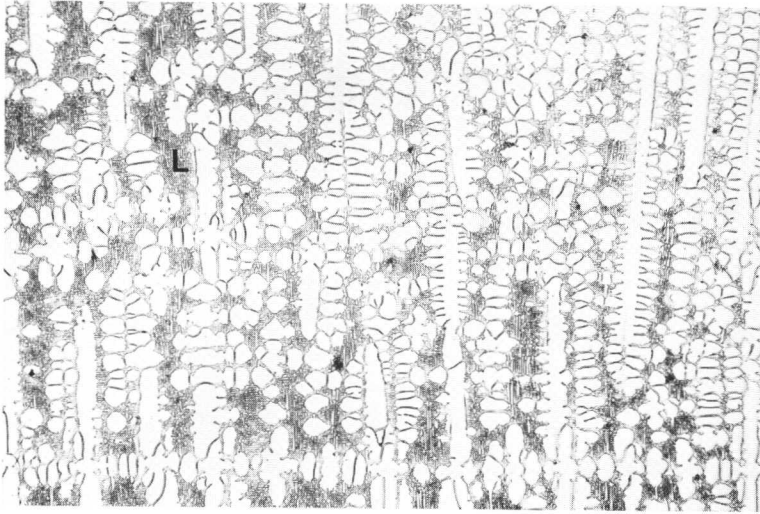
Precipitates

—

Microsegregation

Element	Si	Mn	Cr	Ni	Mo
I	1,8	1,7	1,2	1,1	2,0

$R = 0,5^\circ\text{C/s}$
 $T_q = 1280^\circ\text{C}$



Partly solidified

Figure 1

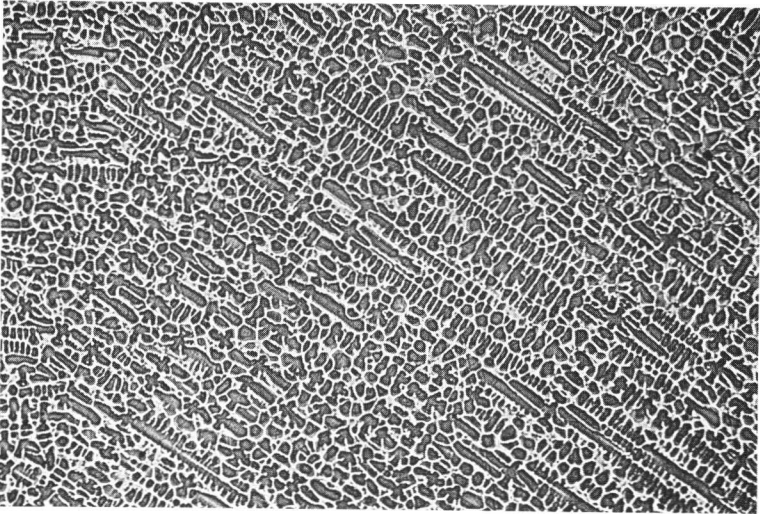
R = 0,5°C/s

T_q = 1385°C

d = 70 μm

γ-dendrites and quenched liquid (L).

400 μm × 25



Completely solidified

Figure 2

R = 2,0°C/s

T_q = 1280°C

d = 55 μm

Figures 2–4: γ-dendrites.
White interdendritic areas.

400 μm × 25

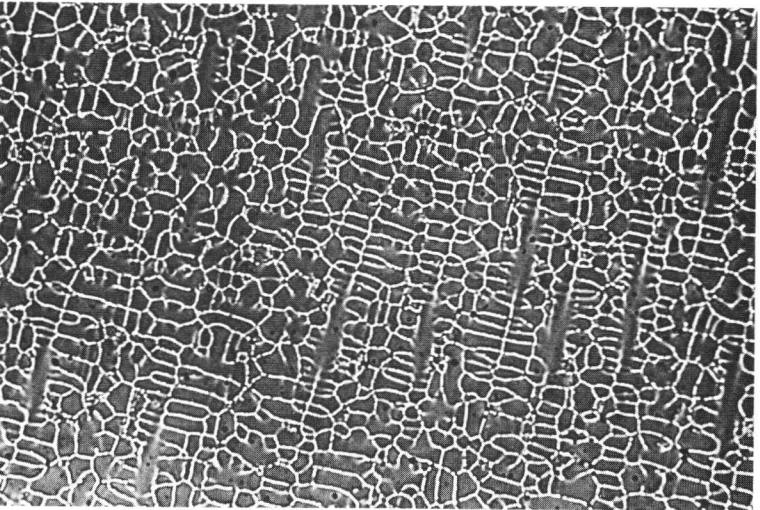


Figure 3

R = 0,5°C/s

T_q = 1280°C

d = 80 μm

400 μm × 25

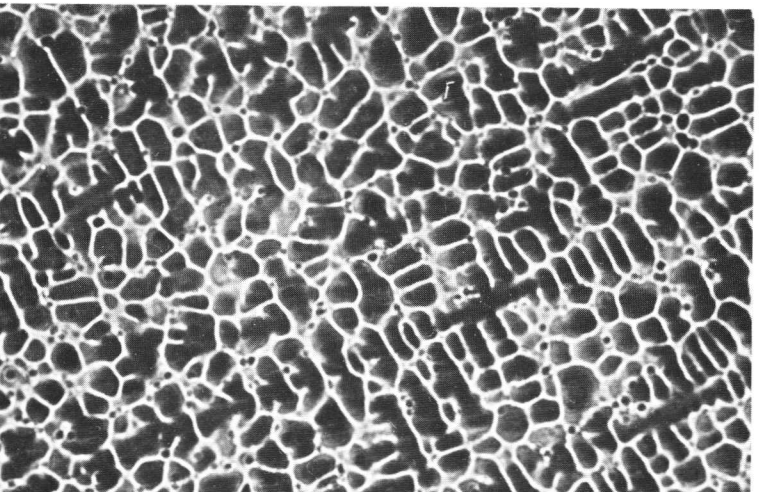


Figure 4

R = 0,1°C/s

T_q = 1280°C

d = 120 μm

400 μm × 25

Figure 5

R = 0,5°C/s

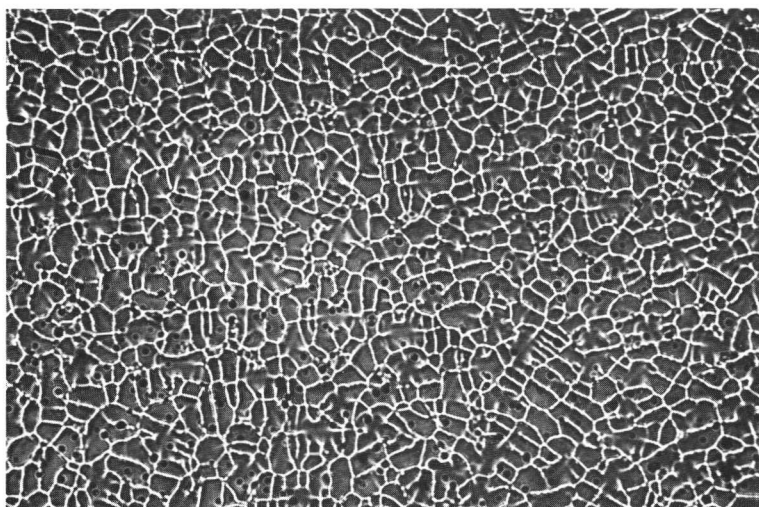
T_q = 1200°C

(d₁₂₀₀ = 90 μm)

γ-dendrites.

White interdendritic areas.

× 25  400 μm



STEEL 414. 0,4 % C 25 % Cr 20 % Ni HEAT RESISTANT STEEL

Designations

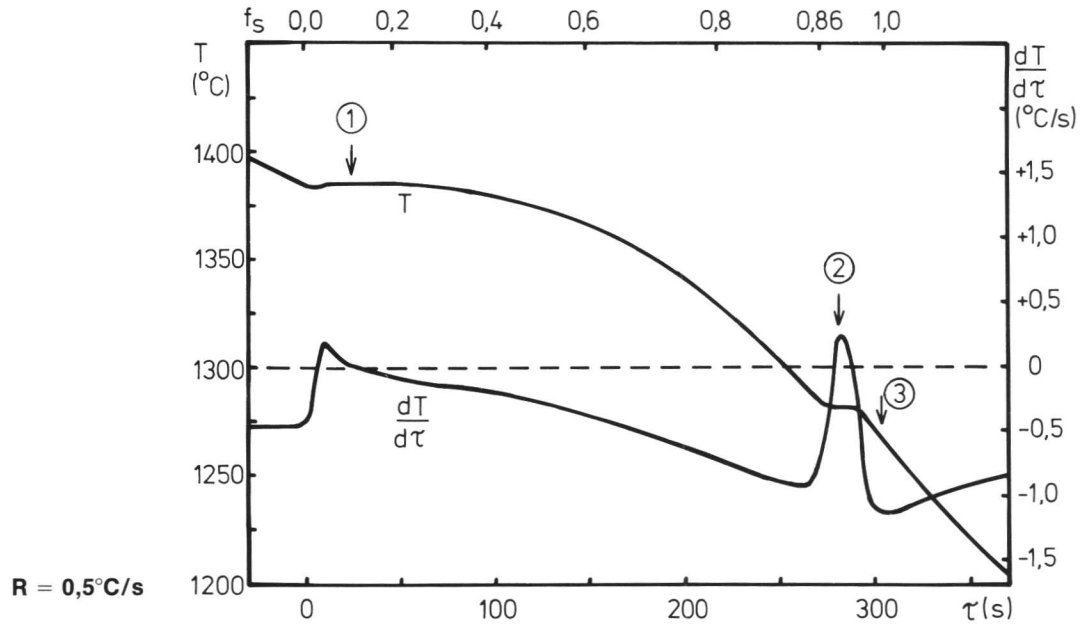
SIS	AISI	Werkstoff Nr
–	310 HC	–

Composition (wt-%)

C	Si	Mn	P	S	Cr	Ni	Mo	Cu	Co	Ti	Al _{tot}	N
0,41	1,00	1,34	0,007	0,010	25,2	20,6	0,08	0,02	0,06	0,10	0,016	0,022

$\frac{Cr_{eq}}{Ni_{eq}} = 0,89$

Thermal Analysis



	Average Cooling Rate, R, (°C/s)		
	2,0	0,5	0,1
Liquidus temperature, austenitic primary phase, °C (1)	1383	1385	1385
Temperature of formation of eutectic, °C (2)	1275–1260	1285–1275	1290–1280
Solidus temperature, °C (3)	1260	1275	1280
Solidification range, °C	125	110	105
Solidification time, s	125	290	1140

Precipitates

Interdendritic M₂₃C₆ – eutectic. The amount of carbide eutectic increased with increasing cooling rate, (see figures 6 – 12).

Microsegregation

Element	Si	Mn	Cr	Ni
I	2,1	1,6	1,2	1,1
P _{ID}	Carbide/γ		1,6	

R = 0,5 °C/s
T_α = 1230 °C

Partly solidified

Figure 1

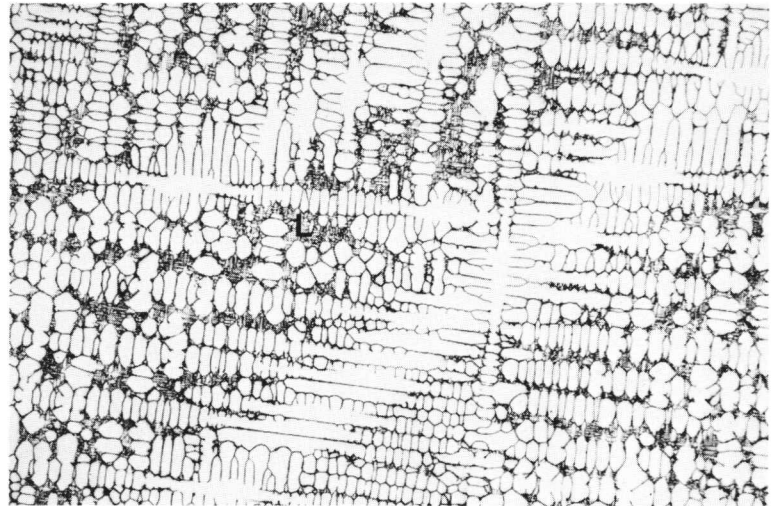
R = 0,5°C/s

T_q = 1375°C

d = 60 μm

γ-dendrites and quenched liquid (L).

× 25 



Completely solidified

Figure 2

R = 2,0°C/s

T_q = 1230°C

d = 50 μm

Figures 2–4: γ-dendrites and interdendritic carbide eutectic, (compare figures 6–10, 12).

× 25 



Figure 3

R = 0,5°C/s

T_q = 1230°C

d = 80 μm

× 25 

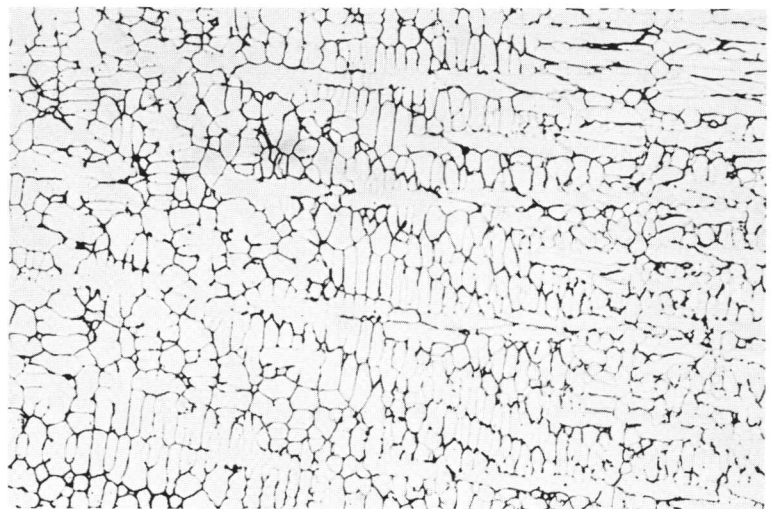


Figure 4

R = 0,1°C/s

T_q = 1230°C

d = 105 μm

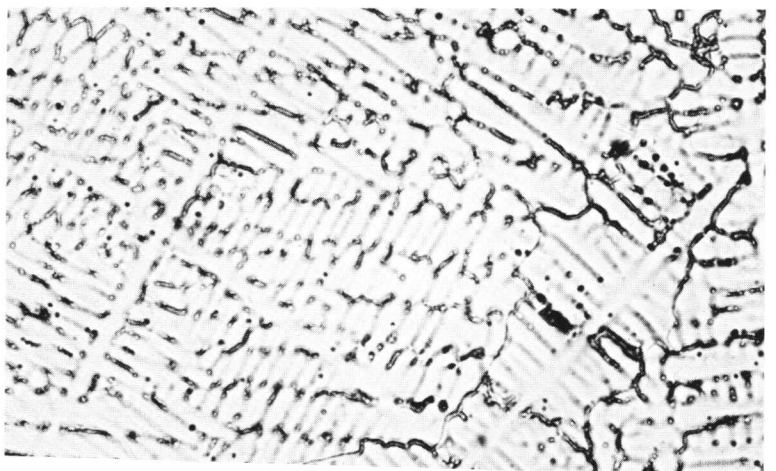




Figure 5

$R = 0,5^{\circ}\text{C/s}$
 $T_q = 1100^{\circ}\text{C}$
($d_{1100} = 90 \mu\text{m}$)
 γ -dendrites and interdendritic carbide eutectic,
(compare figure 11).

400 μm $\times 25$

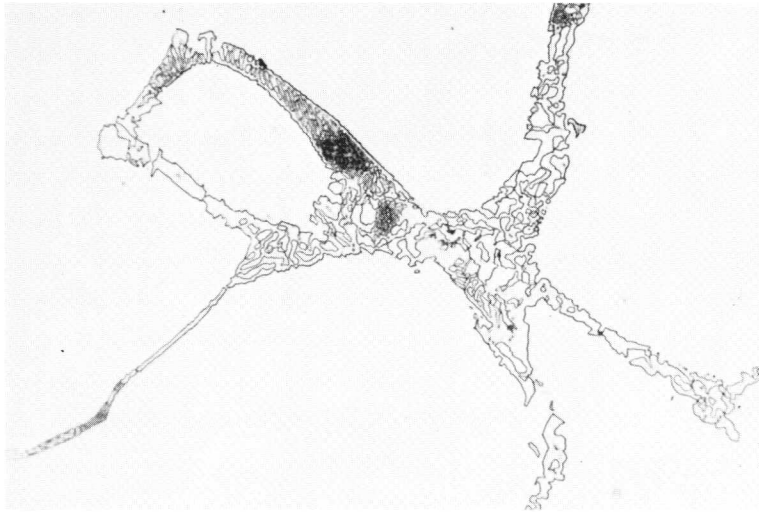


Figure 6

$R = 2,0^{\circ}\text{C/s}$
 $T_q = 1230^{\circ}\text{C}$
 $M_{23}C_6$ - γ eutectic
Figures 6–8: Note the influence of cooling rate
on carbide coarseness.

25 μm $\times 600$

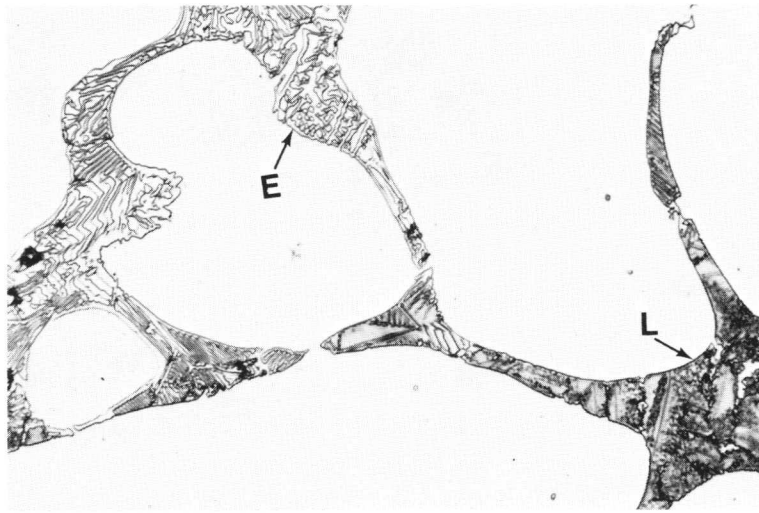


Figure 7

$R = 0,5^{\circ}\text{C/s}$
 $T_q = 1230^{\circ}\text{C}$
 $M_{23}C_6$ - γ eutectic (E)
and residual melt (L).

25 μm $\times 600$

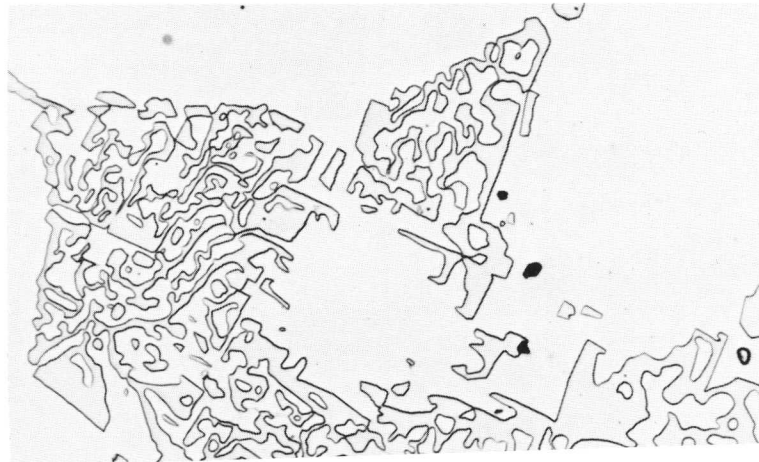


Figure 8

$R = 0,1^{\circ}\text{C/s}$
 $T_q = 1230^{\circ}\text{C}$
 $M_{23}C_6$ - γ eutectic.